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1 Effects of ZnO nanoparticles in the Caspian roach (*Rutilus rutilus caspicus*)

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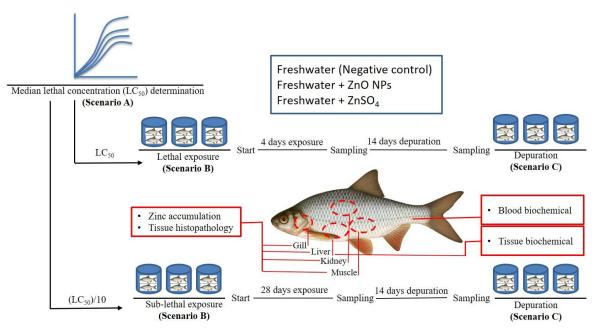
Most studies investigating the toxicity of zinc oxide nanoparticles (ZnO NPs) focused on the effect of size, whereas exposure concentration and duration remained poorly understood. In this study, the effect of acute and sub-acute exposures of ZnO NPs on Zn compartmentalization and biomarkers' expagession were investigated in Rutilus rutilus caspicus (Caspian roach) considering various explosure scenarios: i) the assessment of the concentration-response curves and median lethal configuration (LC₅₀); ii) the assessment of the effects of organisms exposed at LC₅₀ value and one ten**42** of LC₅₀ value of ZnO NPs suspensions for 4 d and 28 d, respectively; iii) the assessment of 14 d depuration period. Same concentrations of ZnSO₄ were investigated. The highest Zn accumulation wasdetected in gill after sub-acute exposure (4.8 mg/L; 28 d) followed by liver, kidney and muscle. In **g**fil, liver and muscle, Zn from Zn NPs accumulated higher concentrations. Depuration (14 d) dec46 ased Zn content in each organ, but no complete removal occurred except for muscle. Biomarkers' activity was significantly over expressed after treatments, but depuration brought back the 48 values to background levels and most effects were related to acute concentrations (48 mg/L; 4 d) 49d in presence of ZnSO₄. Histopathological analyses showed that the exposure to ZnO NPs inc**5**e ased lesions in gill, liver and kidney, with a direct proportionality between alterations and Zn acc51 mulated in the target organs. After depuration, lesions regressed for both ZnO NPs and ZnSO₄, but52 to in a complete way. These data could contribute to increase the knowledge about ZnO NPs risk assessment in aquatic vertebrates, suggesting that the size of ZnO NPs can influence biomarker and histopathological effects.

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Key6vords

Tosicokinetic; acute and sub-acute concentrations; fish; ZnO NPs; ZnSO₄

Grāphical abstract



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1. Introduction

Nafa2technology is one of the most innovative field of the century exploiting physico-chemical profactives of the matter related to its size (~ 10⁻⁹ m) (Minetto et al., 2016; Minetto et al., 2014). Nafa4particles (NPs) are used in a range of household and industry products (Lofrano et al., 2016; Mafa6rer-Jones et al., 2013). Large-scale production and use are likely to result in the release of NPs-bas661 products into the environment making them one of the emerging class of contaminants (Caffregaro et al., 2015; Libralato et al., 2016; Vale et al., 2016). Metal oxide NPs are commonly use681 different fields like sunscreen creams, toothpaste, cosmetics, paint, paper, plastic, textile, cer669nics, medicine, electronics, and wastewater treatment. Zinc oxide NPs (ZnO NPs) are produced for70several industrial and household applications as reported in Table 1s (Supplementary Information).

Th**#**2oxicity of ZnO NPs has been shown to vary according to the exposure dose/concentration, test dur**#**3ion, reference protocol, and considered biological model (Christen and Fent, 2012). ZnO NPs eff**#**4ts towards bacteria (mortality) and marine phytoplankton (growth rate, Miller et al. (2011)), *Ca#borhabditis elegans* (survival, Ma et al., (2009)), zebra fish (mortality and hatching rate, Bai et al. **76**010)), and common carp (oxidative stress and organ-specific accumulation of Zn, Hao and Ch**#**7**1** (2012)) were all dependent on NPs size. Thus, most studies focused on the effect of NPs size, wh**#**8 exposure concentration and duration remained poorly investigated, also, despite importance of **#**9e released Zn ion, just few studies compared the toxicity of Zn-based NPs with ionic Zn in ver**8eb**rates (Hao et al., 2013).

Zhao et al. (2013b) evidenced that NPs could be uptaken and compartmentalized in various tissues geragrating oxidative stress and histopathological damage (Zhao et al., 2013b), but there is a gap into the samowledge about ZnO NPs toxicokinetic including bioavailability, uptake dynamic, tissue disadbution, accumulation, and depuration in organisms. Few studies have examined the bio-disadbution of ZnO NPs in vertebrates. Ates et al. (2015) compared the effects of dietary and wasabborne exposure to ZnO NPs stating that diet can play a major role in Zn bioaccumulation 5

prifizirily via intestine, gills and liver in *Carassius auratus*. Zn bioaccumulation in liver and gill of *CyBBinus carpio* after 21 d exposure to ZnO NPs and ionic form (Zn²⁺) was reported by Hao et al. (2013) showing severe histopathological changes by increasing cellular oxidative stress response; Zn 100 oaccumulation was reported in mice after intravenous injection accumulating preferentially in kid902 y, thigh bone and the gastrointestinal tract (Yeh et al., 2012). Zn from ZnO NPs accumulated also2 in plants like *Zea mays* with significant reduction of root and shoot biomass production (Zhao et all, 2013a). Antioxidant defence responses are commonly used as biomarkers to detect state, sus204 ptibility and exposure to environmental pollutants (Rudneva, 2013). There are limited studies on 916 toxic effects of ZnO NPs on the antioxidant system of vertebrate species (Kaptaner et al., 2016). Plasma glucose and cortisol have been measured as general stress markers in response to different pollutants (Katuli et al., 2014a; Katuli et al., 2014b) and their measurement can help to det98 mine the effects of contaminants on living organisms.

Aquatic environments can be considered as the ultimate sink for many environmental pollutants suctions NPs (Degger et al., 2015; Lenartova et al., 1997), therefore, aquatic wildlife species are inctentsingly at risk. Gottschalk et al. (2009) reported that ZnO NPs forecast concentrations can range2 from 10 and to 430 ng/L in natural surface water and in treated wastewater at European level, respectively. Given their widespread application, it is expected that their environmental levels could furtioner increase in the near future (Osmond and Mccall, 2010). Generally, fish is considered an intendes ting biological model, but data on ZnO NPs are still scarce (Degger et al., 2015; Katuli et al., 201466). Bai et al. (2010) investigated the toxicity of ZnO NPs reporting on the effects on embryos of *Dubnio rerio*. No data about Zn compartmentalization in organs are available, but it showed to be ableous kill embryos (50 and 100 mg/L), retard their hatching (1-25 mg/L) and reducing their body lengt09 and causing tail malformations (after 96 h).

Up**1100**now, many studies have examined the toxicity properties of different NPs on different aspects of **htt** platic organisms' life (Jang et al., 2014; Krysanov et al., 2009; Rajkumar et al., 2016), nevertheless their toxic potential are still not completely understood, and sometimes different 6

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result3 have been obtained mainly due to the differences in exposure methods. While some studies contsidered short-term exposure of acute NPs concentrations (Krysanov et al., 2009), Lee et al. (201115) examined the effects of long-term chronic exposure to NPs. Few studies simultaneously investigated acute and sub-acute effects of NPs (Katuli et al., 2014b). Despite the current use of ZnO1NPs and the increasing possibility of exposure to aquatic organisms to them, there are many uncertainties regarding the potential toxicity of ZnO NPs on different aspects of aquatic organisms espetically towards vertebrates.

This 20 escarch was carried out to evaluate the effect of acute (4 d) and sub-acute (28 d) exposure to Zn 02 NPs considering uptake, Zn accumulation in target tissues and Zn depuration like as the pote 22 ial disruption in the expression of some biomarkers (superoxide dismutase (SOD), catalase (CA23), glutathione-S-transferase (GST), lactate dehydrogenase (LDH), glutathione (GSH), mal 24 dialdehyde (MDA), protein concentrations, blood cortisol and glucose, and tissue his togs athology) in the Caspian roach (*Rutilus rutilus caspicus*). Effects of ZnO NPs were compared to it 26 c Zn (Zn²⁺ from ZnSO₄). To the best of our knowledge, this is the first study investigating the effect Zn of acute and sub-acute concentrations of ZnO NPs on these parameters after depuration period contributing to elucidate the pathway of ZnO NPs toxicity.

2. Magerials and methods

2.11 Experimental design

Experiments were carried out considering: i) the assessment of the concentration-response curves and B2edian lethal concentration (LC₅₀) data after 1, 2, 3 and 4 d of Caspian roach exposure to ZnO NP\$330 uspensions – static renewal acute test (Scenario A); ii) the assessment of the effects of ZnO NP\$330 nd ZnSO₄ in organisms exposed at equivalent zinc concentrations (LC₅₀ value and one tenth of 136_{50} value of ZnO NPs suspensions) for 4 d and 28 d, respectively, looking for Zn compartmentalization and biomarkers' expression (Scenario B); iii) the assessment of 14 d departmentalization period, looking for Zn compartmentalization and biomarkers' expression (Scenario C). Allastenarios were carried out in triplicate according to the Organization for Economic Cooperation and Brevelopment protocol (OECD, 1992). Selected exposure concentrations were in accordance to pre140us results from Hao et al. (2013) on *C. carpio* exposed to ZnO NPs.

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2.21@hemicals

Containercial ZnO particles were purchased from Pishgaman Cop. (Mashhad, Iran). According to the mathefacturer, NPs were spherical with an average size of 30 nm, specific surface area of 60 m²/g (Brittfauer Emmett Teller) and purity of 99.9%. The particle shape and size was characterised statistic from the dry powder and stock suspension by transmission electron microscopy (TEM) (Hitta7hi, Japan) (Figure 1A and 1B) and assessed via ImageJ 1.51j8 (Schneider et al., 2012). ZnSO₄ 7H108 with a purity 99.9% were provided by Sigma- Aldrich (Steinheim, Germany). The stock suspensions were prepared by adding ZnO NPs and ZnSO₄ dry powder into aerated ultra-pure water folltsoed by 30 min of sonication (40 kHz, 100W; USH650, Sonicator, USA) in an ice water bath. Testing suspensions were prepared diluting the stock suspension with aerated tap water (7 mg/L of dis462ved oxygen; pH =7.8 ± 0.1 at 23 ± 1 °C; salinity of 0.25 ± 0.01 mg/L; hardness of 165 ± 8 mg/E30f CaCO₃; SO₄²⁻ of 37 ± 1 mg/L; N-NH₄⁺ of < 0.01 mg/L; N-NO₂⁻ of < 0.01 mg/L; N-NO₃⁻ of 17 ±50.01 mg/L; F⁻ of 0.01 ± 0.01 mg/L; total As, Cd, Cu, Pb, and Zn < 1 µg/L). Before use, tap water55was filtered on activated carbon after ultra-filtration (0.2 µm) The same water was used for Zn**366** solutions.

The 5 \vec{p} article size distribution (hydrodynamic diameter) and zeta (ζ)-potential in the exposure suspeasion were measured by Dynamic Light Scattering (DLS) (Zetasizer, Malvern Instruments).

Tottel Σ n content and dissolution rates were measured after 24 h (for each exposure concentration of Sceneric B). Hydraulic radius distribution of ZnO NPs and zeta (ζ)- potential were analysed by collecting samples from fish water tanks every 24 h (i.e. starting from the zero time) for five days 8

(i.e162) ur half-tank water renewals) considering the 48 mg/L of ZnO NPs exposure concentration. TEM3 analysis of particle size distribution in water samples collected from fish tanks was carried out imf64 liately after dosing.

Airtbebbling facilities present in fish tanks was used to prevent ZnO NPs from settling keeping the suspension constantly mixed. The dissolution rate of ZnO NPs in exposure media was assessed using the Amicon ultra-centrifugal filters (3 kDa, Millipore, Germany). A volume of 15 mL of ZnO NPs 68 uspensions (at 48 and 4.8 mg/L collected after 24 h from dosing from fish tanks) were transferred to the filters and centrifuged at 7168g for 40 min (27 ± 1 °C). The amount of Zn was determined by Inductively Coupled Plasma Mass Spectrometry (ICP-MS, Thermo Finnigan) (Reed et al7,12012) both in the filtrate and in the whole exposure media, and the amount of ion released was 22 xpressed as percentage of the total Zn content. The ICP-MS was calibrated with standard Zn solution (1000 mg/L Zn in nitric acid; Sigma, USA).

2.31Etotoxicity

2.311.50rganisms

Castpian roach specimen with a mean body weight of 16.7 ± 0.76 g and length of 17.3 ± 1.32 cm weight provided from the Sijowal Fish Reproduction Center (Golestan province, Iran). Actilianatization to test conditions prior to exposure consisted in keeping organisms in three tanks of 100079 for 2 weeks under natural photoperiod regime (14: 10 h light: dark) and fed *ad libitum*. All protactions were carried out in accordance with the Animal Care and Use Committee guidelines at the listiculty of Sciences of the University of Tehran (357, 8 November 2000).

Experiments were conducted in 100 L fiberglass tanks that were continuously aerated. Main water paralleleters were kept constant during all the experimental activity: 25 ± 1 °C, 7.0 ± 0.1 mg/L of DOLEAH 7.8 \pm 0.8 165 \pm 8 mg/L of total hardness). Organisms were fed at a rate of 1% body weight per185. The chemical composition of feed for both acclimatization and experiment is shown in Table62S (Supplementary Information). Every 24 h, half of the exposure medium was renewed with freshedy spiked water to remove the metabolic waste of fish within the Scenario A. The experimental design included the analysis of the effects at 0, 10, 20, 40, 50, 80 mg/L of ZnO NPs (nominal contempt tations). Fishes were deprived of feed for 24 h prior to and during the toxicity test. A batch of tt900 rganisms was used during exposure and fish mortality recorded every day for the whole test duration; dead organisms were systematically removed. Effects were reported as LC_{50} and calceleted using the Probit method for each exposure period (Katuli et al., 2014b). According to Scetter in B, fishes were exposed to 48 mg/L of ZnO NPs and ZnSO₄ for 28 d (sub-acute exposure). Within Scenario C, depuration was investigated in triplicate following the same experimental design of Scenario B, except for the fact that after 4 and 28 d of exposure, living organisms were transferred to clean water (free of ZnO NPs and 2nO NPs and

Living specimens, collected at the end of the exposure scenario, were euthanized with an overdose of **1990** mg/L of clove oil for 30 s. After euthanasia, fish gill, liver, kidney and muscle were collected. Tissues were stored at -80 °C for the quantification of total Zn and the analysis of anticondidant enzyme activities, or fixed in 10% buffered formalin solution (Roberts, 2012) and stored at 4 \pm 1 °C until for histopathology. Blood was drawn from caudal vessels and centrifuged (1526) to extract the serum that was stored at -80 °C until further analysis.

Wizłow each exposure scenario, data were tested for normality (Kolmogorov-Smirnov test) and deszon prive statistics. Student's t-test ($\alpha = 0.05$) and one-way analysis of variance (ANOVA) with poszelo C Tukey's test ($\alpha = 0.05$) described the potential significant difference between the conzon dered exposure conditions. The SPSS software version 19.0 (SPSS Inc., Chicago, IL, US) was usezba

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2.322.0Zn compartmentalization

Gil2,11iver, kidney and muscle samples were removed from liquid nitrogen and lyophilized. Ly2philized tissues were separately homogenized with porcelain mortar, then an amount of ~ 0.1 g of 2iksue was transferred to 50 mL digestion vessel containing 5 mL of nitric acid (Merck, Ge2nhany) and 2 mL of hydrogen peroxide (Merck, Germany). Digestion occurred in an oven using a t2nkse-stage protocol: i) 10 min at 150 °C; ii) 15 min at 180 °C; iii) >10 min of cooling phase. Digestied samples were diluted to a final volume of 50 mL using ultra-pure water. ICP-MS analysis wa217arried out to determine the total Zn concentration as well as Zn content in dry-weight target org2nt8 expressed as μ g of Zn per g of dry weight (d.w.) biomass (μ g/g d.w.).

2.321.9Biomarkers and histopathology

Sev222a biomarkers (SOD, CAT, GST, LDH, GSH, MDA, protein content) were assessed in liver tiss222ts. Liver samples were weighed and homogenized at a ratio of 1:10 in sodium phosphatebuf222ed saline (pH 7.3, 300 mmol sucrose, 1 mmol EDTA, and 1.4 mmol dithioerythritol). Sp22232nens were centrifuged at 16128g for 15 min at 4 °C. Supernatants were assessed for activity of \$224D, CAT, GST, LDH, GSH and MDA. SOD, CAT, GST, LDH, GSH, MDA, and proteins we22251etermined according to Winterbourn et al. (1975), Aebi et al. (1984), Habig et al. (1974), Par222Azmon Com. (Tehran, Iran), Tietze (1969), Satoh (1978), and Bradford (1976), respectively. Bl0227 cortisol and glucose were determined according to Katuli et al. (2014b). All specifications ab0228 SOD, CAT, GST, LDH, GSH, MDA and protein determination are available in Suj24Bementary Information (S1).

To 23 extect histopathological markers, gill, kidney, liver and muscle samples were removed from for 23 falin solution and embedded in paraffin blocks and sectioned (4 μ m) using microtome (M2322) dis, 4055) (Roberts, 2012). Sections were stained with haematoxylin and eosin as described by 23 climurugan et al. (2007). Morphological examination was carried out under a light microscope and pactures were captured using Nikon EC 600 Eclipse microscope.

3. Results

3.12 Rarticle characterization, ZnO NPs suspension stability, and Zn dissolution rate

Zn**28**NPs characterization is shown in Figure 1. According to TEM analysis, primary particle size of **dBg** powder ZnO NPs ranged between 20-70 nm (mean = 45 ± 15 nm, median = 55, n = 100) (FigBBe 1A), while particle size distribution in water samples collected from fish tanks (with fishes) im**2**Accliately after dosing (48 mg/L) ranged between 40-130 nm (mean = 90 ± 16 nm, median = 85, n = **240**0) (Figure 1B).

Hyddaulic radius of ZnO NPs present in water samples collected from fish tanks (with fishes) imagestiately (0 h) after dosing (48 mg/L) were presented in Figure 1C ranging between 37-141 nm $(mean = 79.3 \pm 15.8 \text{ nm}, \text{median} = 73 \text{ nm}, \text{ at } 25 \text{ °C})$. Hydraulic radius of ZnO NPs present in water san2ples collected from fish tanks (with fishes) after 24 h from dosing (48 mg/L) were presented in Fig246 1D ranging between 58-342 nm (mean = 163 ± 37 nm, median = 141 nm, at 25 °C). The inczetase of the average values of hydraulic radius indicated aggregation/agglomeration with an inczease of 48% after 24 h. After fish tank water renewal and 1 h equilibration post-renewal each tim249 or five days, the hydraulic radiuses still indicated aggregation/agglomeration. Differences ran**25d** up to 33%, but they were not significantly different (p < 0.05) than hydraulic radius dis25bution after 24 h. The zeta-potential of ZnO NPs in water samples collected from fish tanks (wi262 fishes) was -17.545 ± 7.27 mV and -21.46 ± 4.81 mV at 0 h and after 24 h from dosing (pH = 7.72530.1), in that order. The ZnO NPs suspension was relatively stable potentially supporting slight agase agase ation/agglomeration according to Patel and Agrawal (2011) as also suggested by DLS analysis. The zeta-potential of not spiked tap water was 2.55 ± 1.07 mV (pH = 7.6 ± 0.1). After fish tank5% ater renewal and 1 h equilibration post-renewal each time for five days, zeta-potential did not significantly differ than after 24 h with values suggesting a trend towards slight/medium agassation/agglomeration (up to 28% more than after 24 h). These data suggested that testing sus**259** sions were relatively stable during the exposure time.

The **60**Ps dissolution rates were assessed in water samples collected from fish tanks (with fishes) at 48 **260** to exposure) and 4.8 (sub-acute exposure) mg/L of ZnO NPs. The amount of Zn^{2+} released as **p60** centage of the total Zn content was $4 \pm 2\%$ and $7 \pm 5\%$ for the sub-acute and acute exposure, res**p68** tively. They showed to be statistically independent from the starting exposure concentration (p **260**.05), and relatively constant during the exposure period (i.e. first five exposure days and last expressive day) (up to a maximum of 28% variation).

3.226 cenario A: acute toxicity of ZnO NPs suspensions and ZnSO₄ solutions

Th**26** nortality of Caspian roach in acute exposure assays (Scenario A) showed a concentration-time departality of Caspian roach in acute exposure assays (Scenario A) showed a concentration-time departality of Caspian roach in acute exposure controls, no mortality effects were highlighted during the **269** posure period (up to 4 d). Toxicity as LC₅₀ (and relative 95% confidence limit values) for Zn**07** NPs suspension increased over time: 24 h (78 ± 7 mg/L), 48 h (61 ± 5 mg/L), 72 h (53 ± 6), and **279** 6 h (48 ± 3 mg/L). This information was used for the sub-sequent exposure scenarios (Sc**277** ario B and C) also to test ZnSO₄. For acute and sub-acute exposures, real concentrations of Zn in **Zn9** NPs suspension were 34.1 ± 6.1 and 2.9 ± 0.8 mg/L, and in ZnSO₄ solutions were $26.78 \pm 7.22374d$ 3.14 ± 1.82 mg/L, respectively.

3.32 Scenario B and C

3.321 Partitioning of Zn in target organs

The **ZD** ncentration of Zn in target organs was summarised in Figure 3 as total Zn content ($\mu g/g \, d.w.$) in **gifts** liver, kidney and muscle after 4 d (acute exposure, 48 mg/L of ZnO NPs and ZnSO₄) and 28 d (**aub**-acute exposure, 4.8 mg/L of ZnO NPs and ZnSO₄) of contact time (Scenario B), and 14 d deparation (Scenario C). Results showed that all tissues significantly (p < 0.05) increased their total Zn **28** incentration compared to negative controls in both exposure conditions and in both Scenario B and **82** The only exception was for muscle after depuration. Also, the comparison between Scenario

B **264** C in each Zn forms and concentrations (identified by asterisk) showed that Zn had a sig**264** cant reduction in Scenario C except for the acute concentration of ZnSO₄ in gill and muscle. Fo2**265**th ZnO NPs and ZnSO₄, bioaccumulation trends of Zn were similar in gill, kidney (p < 0.05) and **266** re with higher Zn levels after sub-acute exposure (p < 0.05). In muscle, Zn content did not sig**267** cantly change (p < 0.05) between acute and sub-acute exposure, but decreased to background lev**288** (not significantly different from negative controls, p < 0.05) after depuration. In kidney, Zn fro**286**/2nSO₄ suspension presented the highest concentration, while in other tissues Zn from ZnO NP**290** as prevalent.

The grartitioning of Zn into the target organs of non-exposed organisms (negative controls) was: i) acuae ($14 \pm 1 \mu g/g d.w.$), kidney ($36 \pm 7 \mu g/g d.w.$), liver ($67 \pm 18 \mu g/g d.w.$), and gill $(572933 \ \mu g/g \ d.w.);$ ii) sub-acute exposure (28 d): muscle (14 ± 1 \ \mu g/g \ d.w.), kidney (49 ± 21 \ \mu g/g) d.w294 liver (49 \pm 32 µg/g d.w.), and gill (46 \pm 8 µg/g d.w.). Data from control groups were averaged within the target organ (being not statistically different, p < 0.05) and used to define the Zn 296 kground level used in Figure 3. In summary, Zn partitioned similarly after the two exposure per29d s (not considering the 14 d depuration) within Scenario B: i) acute exposure (4 d at 48 mg/L): mu20 Re $(34 \pm 1.5 \ \mu g/g \ d.w.$ for Zn from ZnO NPs; $19 \pm 2 \ \mu g/g \ d.w.$ for Zn from ZnSO₄), kidney $(14299 \pm 157 \text{ d.w. for Zn from ZnO NPs}; 1817 \pm 123 \mu g/g \text{ d.w. for Zn from ZnSO}_4)$, liver $(2123 \pm 123 \mu g/g \text{ d.w. for Zn from ZnSO}_4)$ 24300bw. for Zn from ZnO NPs; $1749 \pm 176 \ \mu g/g \ d.w.$ for Zn from ZnSO₄) and gill (2541 ± 127) d.w20fbr Zn from ZnO NPs; $1226 \pm 174 \,\mu\text{g/g}$ d.w. for Zn from ZnSO₄); ii) sub-acute exposure (28 d at 402 mg/L): muscle (38 ± 3 d.w. for Zn from ZnO NPs; 18 ± 2 µg/g d.w. for Zn from ZnSO₄), kidB03y (1922 \pm 237 d.w. for Zn from ZnO NPs; 2450 $\pm \mu g/g$ d.w. for Zn from ZnSO₄), liver (2314 \pm 1304d.w. for Zn from ZnO NPs; 2453 \pm 372 µg/g d.w. for Zn from ZnSO₄) and, gill (3036 \pm 221 d.w20for Zn from ZnO NPs; $2012 \pm 176 \,\mu\text{g/g}$ d.w. for Zn from ZnSO₄). The highest Zn content was det80fed in gill after 28 d after sub-acute exposure to ZnO NPs and the lowest was shown in muscle (wi3b7slight differences between the two exposure scenarios) (data are presented in Supplementary DaßOFFigure 1s A).

Aft309the 14 d depuration (Scenario C), the partitioning of Zn into the target organs of not exposed organities (negative controls) was: i) acute (4 d): muscle ($15 \pm 1.5 \ \mu g/g \ d.w.$), kidney ($15 \pm 7 \ \mu g/g$ d.w3), there (74 ± 29 μ g/g d.w.) and gill (68 ± 9 μ g/g d.w.); ii) sub-acute exposure (28 d): muscle (9 \pm (B52µg/g d.w.), kidney (61 \pm 10 µg/g d.w.), liver (53 \pm 17 µg/g d.w.) and gill (68 \pm 3 µg/g d.w.). DaBal from control groups were averaged within the target organ (being not statistically different, p < (B05) and used to define the Zn background level used in Figure 3. After the 14 d deputation (Scatizirio C), the partitioning of Zn into the target organs of exposed organisms was: i) acute expansive (4 d at 48 mg/L): muscle (14 \pm 3 for Zn from ZnO NPs; 13 \pm 1.5 µg/g d.w. for Zn from ZnSO₄), kidney (479 ± 47 for Zn from ZnO NPs; 521 ± 84 μ g/g d.w. for Zn from ZnSO₄), liver (15118±58 for Zn from ZnO NPs; 978 ± 121 μ g/g d.w. for Zn from ZnSO₄) and gill (708 ± 152 for Zn 310 MPs; $727 \pm 186 \ \mu g/g \ d.w.$ for Zn from ZnSO₄); ii) sub-acute exposure (28 d at 4.8 mg320: muscle (15 \pm 2.5 for Zn from ZnO NPs; 11 \pm 1 μ g/g d.w. for Zn from ZnSO₄), kidney (856 \pm 83/2 for Zn from ZnO NPs; 684 \pm 176 µg/g d.w. for Zn from ZnSO₄), liver (1721 \pm 54 for Zn from Zn Θ 2NPs; 1384 ± 89 (µg/g d.w. for Zn from ZnSO₄) and gill (1716 ± 160 for Zn from ZnO NPs; 750328 236 µg/g d.w. for Zn from ZnSO₄). After 14 d depuration period, highest Zn content was deta2red in liver in sub-acute concentration of ZnO NPs and the lowest was still shown in muscle (wi3b25slight differences between exposure scenarios) (data are presented in Supplementary Data Fig326 1s B).

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3.3328Liver biomarkers

Results from the analysis of biomarkers (SOD, CAT, GST, LDH, GSH and MDA) were presented in **BBQ**ure 4. Results from negative controls did not significantly differ between exposure conditions (acBB4 and sub-acute) for both ZnO NPs and ZnSO₄, so mean activity values were used for negative controls (i.e. a mean value for acute and sub-acute exposure, and a mean value after depuration incB33 ing results after 4 d and 28 d exposure). All3Bitomarkers were significantly overexpressed compared to negative controls (p < 0.05) for both botb33ZnO NPs and ZnSO₄, and acute and sub-acute exposures with the except for GSH activity, and LD43Gactivity exposed to the sub-acute concentration of Zn, and GST activity exposed to the subacute 37 concentration of ZnSO₄. CAT, GST, LDH and GSH activity were higher for the acute exp338ure than the sub-acute one (only changes in the GST activity was significant in ZnO NPs susp38sion); while the activities for SOD and MDA in sub-acute concentration was higher (only changes in the SOD activity was significant in ZnSO₄ suspension) (Scenario B). Within Scenario B, ZnG4NPs affected more GST (p < 0.05) and LDH, while ZnSO₄ affected more SOD, CAT (p < 0.0342 and MDA.

After 4314 d depuration (Scenario C), there was no significant differences between negative controls and 441 biomarkers (except for MDA at acute concentration of ZnSO4). Also the activity of SOD, CA314,5GST (except for 4.8 mg/L for ZnSO4), LDH (except for sub-acute concentration), and MDA (ex346pt for 48 mg/L of ZnSO4) significantly decreased compared to the levels in Scenario B (identified by asterisk).

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3.334 Cortisol and glucose in blood

Comparison and glucose results were presented in Figure 5. Results from negative controls did not significantly differ between exposure conditions and mean activity values were used for negative compareds. After exposure to both ZnO NPs and ZnSO₄, cortisol and glucose levels significantly incresses compared to control groups (p < 0.05). The comparison exposures (Scenario B) showed that the magnetic date of ZnO NPs and ZnSO₄ (acute) cortisol significantly augmented compared to the to the sub-acute one (p < 0.05). For glucose, the highest level was showed after 28 d at 4.8 mg/L of ZnSO₄. After the 14 d depuration (Scenario C), both cortisol and glucose concentrations were back to the state pre-exposure levels (i.e. not significantly different from negative controls, p < 0.05), demonstrating that their alteration was reversible. Also, the comparison between Scenario B and C

for $3b\theta$ th ZnO NPs and ZnSO₄ (identified by asterisk) showed that cortisol and glucose had a significant reduction in Scenario C.

3.336Histopathological effects

Histopathological changes in gill, liver and kidney of Caspian roach within acute and sub-acute expansive scenarios were summarized in Figure 6 and Figure 2s (in Supplementary Materials) and also64 Table 1, 2 and 3. Microscopy observations (Figures 6 and 2s) revealed alterations in the targe5organs after the exposure of organisms to ZnO NPs and ZnSO₄ as from Scenario B and C. Ma366histopathological changes in gill included: shortening of secondary lamellae, collapse of sec3667dary lamellae, curling of secondary lamellae, epithelial lifting, epithelial hyperplasia and langedar fusion (Figure 6A, B, C, Figure 2s A and B and Table 1). Nuclear conjunction, hypegrophy of hepatocytes, hepatic lipolysis and focal necrosis were the most frequent lesions observed in Caspian roach liver after exposure to different concentrations of ZnO NPs and ZnSO₄ (Figute 6D, E, F, Figure 2s C and D and Table 2). Acute and sub-acute concentrations of ZnO NPs and 72 nSO₄ produced lower impacts on kidney, compared to other tissues, but evidencing the desizeration of Bowman's capsule, glomerulus and renal tubule (Figure 6 G, H, I, Figure 2s E and F and Table 3). No changes of muscle structural tissues were found after exposure to different consentration of ZnO NPs (un-published data), probably because it was the less target organ about Zn 3b Baccumulation. After 14 d depuration (Scenario C), most histopathological changes were subsite ntially recovered (Figures 6 and 2s, and also Table 1, 2 and 3). Depuration was less effective in BZS vering gill tissues especially in fish after sub-acute exposure (4.8 mg/L of ZnO NPs for 28 d) (Table 1).

4. Bascussion

4.1382n partitioning and effects

Zn 382cumulation in tissues of aquatic animals showed to affect their structure and function (Kr383anov et al., 2009). Few studies assessed the partitioning and effects of ZnO NPs in fish (Ka8a4koc et al., 2003; Shukla et al., 2007), thus some comparisons with other biological models and NP385ere introduced into the discussion.

Gill&66are the most important organs involved in breathing (Cengiz, 2006) having also other physicological roles such as metabolites excretion, ion exchange and regulation of acid-base balance (Boxega and Lock, 1991). Results from this study showed that the highest Zn accumulation was obsace of a gigas and Lock, 1991). Results from this study showed that the highest Zn accumulation was obsace gigas after 4 d exposure to ZnO NPs, Zn preferentially accumulated in gill and then in these transferse gigas after 4 d exposure to ZnO NPs, Zn preferentially accumulated in gill and then in these organe. Similar results, but with other engineered nanomaterials were presented by Krysanov et al. (2000) with *Poecilia reticulata* that evidenced after 5 d exposure to single walled carbon nanotubes the higher ferential accumulation in gill (i.e. gill > spleen > liver > gonad). Due to their vulnerable extands and primary contact with suspended contaminants and their large surface area, gillsocan be considered at high risk. In previous studies, alterations in gill such as mucus secretion and some perplasia can increase the likelihood of NPs and other chemicals of sticking onto the same targeorgna (Spry et al., 1988; Smith et al., 2007; Bilberg et al., 2010).

Liv**39**9is another critical organ due to its involvement in detoxification processes (Haschek and Ro**496**eaux, 2013). From this study, the accumulation trend of Zn in both acute and sub-acute exp**406**ures is like muscle with no significant difference between acute and sub-acute concentrations. (Fi**g02**e 3). Moreover, there was little difference between ZnO NPs and ZnSO₄ exposure about Zn acc**406**ulation in liver. In Krishnaraj et al. (2016), after 14 d exposure to sub-acute concentrations of Ag**4NP**s, Ag accumulated in liver of adult zebrafish. In study of Krysanov et al. (2009), guppy (*P. ret406ata*) were exposed to sub-acute concentration of SnO₂ NPs with Sn accumulating pre**406**entially in liver. Zinc can reach the liver passing through the gills and via gut absorption then

usi**4**<u>0</u>7blood circulation (Wang et al., 2011). Thus, this could be the potential reason of high Zn con**408**t in Caspian roach liver.

Kidaooy is an organ involved in the metabolism of excreta such as ammonia and creatinine (Katuli et al.,42014a) and acts as a filter in fish for particles present in the blood stream. Results showed for both 12 no NPs and ZnSO₄ that total Zn accumulated more easily in sub-acute concentrations with the4 highest Zn concentrations of accumulated Zn in relation to ZnSO₄. Karakoc et al. (2003) sho446 that after 30 d exposure, Zn accumulated in kidney> gill > liver of *O. niloticus* with levels directed y proportional to exposure time. Al-Bairut et al. (2013) compared the toxicological effects of Cu4NPs and CuSO₄ concluding that in some tissues, CuSO₄was more bioavailable than the nanoform 2nSO₄ accumulated in the present case study potentially explaining the high levels of Zn 4tom ZnSO₄ accumulated in kydney.

Actored ing to the present paper results, there was no relation between Zn accumulation in kidney Zn 4439 cumulation in muscle was lower than all other investigated tissues suggesting that it cannot store 2021 probably due to the lack of metal binding proteins. No significant differences were found bet4424 en acute and sub-acute exposures of Zn considering the Zn levels accumulated in muscle. Raj429 mar et al. (2016) observed the accumulation of Ag in muscle after 7 d exposure to *Labeo rol4424* (10, 25, 50 and 100 mg/L) of Ag NPs. Similarly, Jang et al. (2014) showed that after 7 d exposure to 0.62 mg/L Ag NPs, lowest concentrations of Ag were observed in muscle and brain of cor4255 on carp.

Eli**426** ation of Zn from tissues of Caspian roach was observed after 14 d depuration. Zn acc**427** ulated in gill, liver, kidney and muscle in both acute and sub-acute exposures decreased cor**428** ared to the relative treatments (Figure 3). In summary, the following Zn level reduction per**429** atages were observed for gill, liver, kidney and muscle, in that order: i) acute exposure to ZnO NP**430** nd ZnSO₄: 72% and 41%, 29% and 44%, 66% and 71%, 59% and 31%; ii) sub-acute exp**438** and 63%, 26% and 44%, 55% and 72%, 61% and 39%. Zn red**438** ion trend in gill, liver and kidney was proportional to the relative treatment concentration, 19

bei**43**3higher after the sub-acute exposure. Zn after acute exposure seemed more "labile" and more effi**436**ntly eliminated during the depuration phase. Zn from ZnSO₄ presented the highest acc**486**fulation in kidney, but a significant reduction after depuration as well even if background lev**4B**6(i.e. negative controls) were not reached after 14 days. In muscle, Zn concentration was back to **ba37**kground levels after depuration.

4.24 Effect of ZnO NPs on oxidative stress

The 39 ntioxidant defence system prevents the occurrence of oxidative damage caused by reactive ox years species (ROS), and could be remarkably increased under different stress conditions (Liadley, 1998). Results from Caspian roach exposure to acute and sub-acute concentrations of ZnO NP442nd ZnSO₄ were reported in Figure 4.

SOLDES the first antioxidant enzyme against oxidative toxicity catalysing dismutation of highly supertaxide radical O^{2-} to O_2 and H_2O_2 (Panda, 2012). SOD significantly increased than the control groups (p < 0.05) but no significant differences were present between results after acute and sub-acute6exposures. This result may be due to the high concentrations of NPs and self-scavenging capateity of SOD. And also in groups that exposed with ZnSO₄, SOD activity significantly increased in 4440-acute concentration. Muralisankar et al. (2014) observed that the freshwater prawn *Mattabachium rosenbergii* exposed to ZnO NPs, increased its SOD activity in a dose-dependent mattate to Ag NPs augmented in a significant way SOD activity. Conversely, Hao and Chen (2012) repatied that *C. carpio* exposed to ZnO NPs (0, 0.5, 5, and 50 mg/L) after an initial SOD increase (0.356 g/L) it significantly decreased (more than 80%) in a time-concentration manner.

CA45,4located in peroxisomes, facilitates the removal of H_2O_2 (namely the product of SOD activity) that 55 metabolized to molecular oxygen and water (Lindley, 1998). CAT activity significantly inc456 in exposed animals thus it prevented greater oxidative damage in the exposed fish. Ind457 ion of CAT as a response to NPs exposure was previously reported in Cyprinidae (Gül et al.,

20045) Scrobicularia plana (Buffet et al., 2011), *H. diversicolor* (Buffet et al., 2011; Cozzari et al., 20145) and *M. rosenbergii* (Muralisankar et al., 2014).

GS46by using GSH increases dismutation of hydrogen peroxide (Habig et al., 1974; Völker et al., 20145) and can bind Zn to thiol group of GST leading to detoxification, and thus increasing exposed ania62 is tolerance and survival in critical situations (Gül et al., 2004; Yuan et al., 2016). GST active increased after exposure to acute and sub-acute concentrations of ZnO NPs, with the highest levels reached after the acute scenario (Figure 5). Also, comparison between exposed grot65 showed that the lowest activity was related to sub-acute exposure to ZnSO4. According to Bro4666 et al. (2000), Zn can bind to thiol groups (Brown et al., 2000) supporting detoxification, and thu4665ST reduction after 28 d. In previous studies, GST increased in clams exposed to Ag NPs (Väl62er et al., 2015), similarly to *Mytilus galloprovincialis* exposed to CuO NPs after 15 d (Gomes et al.62013).

LD470 catalysing the conversion of lactate to pyruvic acid and back, followed the GST trend with high71 activity after acute exposure than sub-acute one (Figure 4) suggesting the presence of non-spet72 c injuries like cell death (Agrahari et al., 2007). It can be said that Zn caused cell death in expt732 c fish. However to better understand toxicity mechanism of NPs on LDH activity, further inv4374 gation are needed. Ates et al. (2016) and Lee et al. (2014) found that rainbow trout and cor475 on carp exposed to Fe NPs and ZnO NPs, respectively, presented levels of LDH activity sig476 cantly increased compared to the control groups.

GS477s a non-enzymatic antioxidant acting as a protective agent against numerous toxic substances and 72a talysing hydrogen peroxide (Hao and Chen, 2012). GSH activity in exposed groups did not show 9 any significant difference compared to negative controls (Figure 4). Hao and Chen (2012) higherighted that *C. carpio* exposed for 14 d to ZnO NPs presented decreasing GSH activity. Conserved, Völker et al. (2015) showed that after 28 d exposure to Ag NPs, GSH activity increased. Masella et al. (2005) and Nordberg and Arner (2001) evidenced that high GSH values can483crease lipid peroxidation (LPO) disturb the antioxidant balance. The end products of LPO are rea484/e aldehydes, such as MDA.

MD485like GSH is another non-enzymatic antioxidant and has been used extensively as a biomarker of 4866dative stress (Xiong et al., 2011; Zhao et al., 2013b). MDA activity in the Caspian roach sig4877cantly increased in both of acute and sub-acute concentrations (p < 0.05) in a similar way ind4688ting Zn oxidative effects. Significant increase in MDA activity was detected in *D. rerio* exp4899ed to ZnO NPs (Zhao et al., 2013b) and to TiO₂ NPs (Xiong et al., 2011), but Gül et al. (2004) rep4990ed a decreasing MDA trend for Cyprinidae, suggesting the presence of species specific related fac490ts, besides of exposure protocols and type of NPs.

After 214 d depuration, results showed that almost all antioxidant biomarkers significantly decreased compared to treatments being the differences with negative controls not significant suggesting the reverse billity in the Zn related metabolism. Nevertheless, after the depuration period biomarkers were sensitive to background levels, Zn content in gill, kidney, and liver was significantly greater than negative controls and this could be explained with the self-scavenging capacity of antioxidant deferrer systems or increased adaptation in response to the new condition, to be further investigated. In 498 mary, with the exception of GSH, all biomarkers responded to the presence of Zn; most biomarkers were more sensitive to long term exposure 4.3. Blood biochemical factors

CofioCol and glucose are two important general stress markers used to in various studies (Ates et al., 20 56);1 Katuli et al., 2014b). Under hypothalamus-pituitary-internal (HPI) axis stimulation, cortisol sec5602ed from the teleost head kidney increases the energy availability during stress conditions (Katodi et al., 2014a). Results showed that cortisol and glucose levels significantly increased confiquented to negative controls (Figure 5). Similarly, Katuli et al. (2014b) after exposing *D. rerio* to acutie05and sub-acute concentrations of Ag NPs, cortisol and glucose levels significantly increased that for Murray (2016) in the case of rainbow trout exposed to Ag NPs. In this study, configible levels increased more after acute exposure than sub-acute one being probably due to the

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adațateve response of cortisol (Fast et al., 2008) that tends to reduce after long-term exposure to stresse9condition thus gradually reducing its sensitivity. Fast et al. (2008) showed that after short-termstateress cortisol levels significantly increased compared to negative controls and long-term stress grosses. Glucose levels significantly increased in the exposed groups to ZnO NPs and this is related to 502 isol role in gluconeogenesis process resulting in glucose production (Saravanan et al., 2011; Sheridan, 1989). Also Ates et al. (2016) and Lee et al. (2014) reported an increase of glucose levels in fish exposed to Fe NPs (*O. niloticus*) and ZnO NPs (*C. carpio*), respectively. Highest glucose levels is showed after 28 d exposure to sub-acute concentration of ZnSO₄. It can be said that at the highteoncentrations, both forms had equal impact on glucose, but at low concentrations, ZnSO₄ had moset refrects. Similarly, Massarsky et al. (2013) reported the same effect for Ag NPs and AgNO₃ abcute hatching delay in *D. rerio*.

After 1914 d depuration, results showed that cortisol and glucose levels were back to the initial ones wite 2010 statistical difference compared to control groups (Figure 5). Katuli et al. (2014a) showed that 52 fter 96 h recovery period, cortisol and glucose levels significantly decreased after exposure of Caspian roach to diazinon. This suggests that the effects of ZnO NPs and ZnSO₄ on cortisol and glue 52 can be really like other toxic agents and that recovery is possible after the removal of polsatants.

4.451Bistopathological alteration

Th**62f**anel of potential histopathological alterations can be an interesting tool to evaluate the effects of **52b5x** incE333 in the distance and time for contaminants to reach blood stream. As well as gill hyperplasia cau5334 a decrease in gill surface area and a subsequent increase in the toxicant-blood diffusion disE336 (Cengiz, 2006; Katuli et al., 2014a). As mentioned previously (section 4.1) these lesions car536 possible reason for high concentration of Zn in gill. The hyperaccumulation of Zn in gill (FigBa7e 3) damaged its structure and the increased mucus production could have cause an increased gill5636 ament capacity to absorb NPs (Bilberg et al., 2010) contributing to the general disruption of gill536 tivity. Similarly, Rajkumar et al. (2016), Griffitt et al. (2007) and Smith et al. (2007) exposed *L.* 5406 *ita, D. rerio, Oncorhynchus mykiss* to Ag NPs, CuO NPs and SWCNT, respectively, evillet for several lesions in gills.

Live#2 plays an important role in metabolism and detoxification of pollutants (Haschek and RoE4Seaux, 2013) and its pathology is used as an indicator of health status alteration (Federici et al., 200574 Rajkumar et al., 2016). After gill, liver was the second organ that hyperaccumulated Zn (Figue) and presented some visible lesions like nuclear conjunction and hypertrophy of hepatocytes (Figue 6, D-F and Figure 2s C-D; Table 2). Similarly, *L. rohita* exposed to Ag NPs presented liver lesions like formation of vacuolation and vacuolar degenerations (Rajkumar et al., 2016), while Fedue#ci et al. (2007) reported loss of sinusoid space and lipolysis in *O. mykiss* exposed to TiO₂ NPS42and Lee et al. (2012) observed hyperplasia, cytoplasm vacuolation in *C. carpio* after exposure to AgONPs.

Kidstedy is an important organ to keep organism homeostasis via ions exchange and secretion of metablolic products and water and its impairment can damage fish physiology and ultimately surststeal (Katuli et al., 2014a). Several lesions such as degeneration of glomerulus and Bowman's captionary were observed in kidney after exposure to ZnO NPs (Figure 6, G-I and Figure 2s E-F; Table 3) **5155** in relation to the significant Zn accumulation (Figure 3). Similarly, Lee et al. (2014) showed *C.* **556***pio* exposed for 12 weeks to ZnO NPs presented several lesions in kidney like number and shafter of the lysosomes and the renal tubule.

In **fise** case of muscle, no histopathological alterations were found after both acute and sub-acute exp**559**ure to zinc. This result was in accordance to Zn partitioning and accumulation in Caspian roa**56**0 so that there could be a proportionatility between the amount of Zn accumulated in tissue and the5**61**esence of lesions. Overall, it was observed that more histopathological alterations were pre**562**t after sub-acute exposure (28 d at 4.8 mg/L of ZnO NPs).

Afts 6314 d depuration, fewer lesions were observable and tissue health generally improved. Most les to the high amount of Zn stored in this organ even after departs tion.

5. 666 clusion

Restricts from this study could help modelling ZnO NPs effects and compartmentalization mestrianisms suggesting the need for future research considering environmentally relevant conficted need as assessing the environmental risks associated to the release of ZnO NPs intersection and a suggesting the environmental risks associated to the release of ZnO NPs intersection and the environment also considering the potential mixture effect in presence of other confitation intersection.

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De6Baration of interest

The sathers report no conflicts of interest related to this study.

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Table 1. Summary of histopathological effects in gill of *R. rutilus caspicus* after acute (48 mg/L, 4 d) and sub-acute (4.8 mg/L, 28 d) exposure to ZnO NPs and ZnSO₄ (Scenario B) and subsequent depuration (14 d) (Scenario C).

		Lesions			
Exposure scenario	Duration	Shortening of secondary lamellae	Collapse of secondary lamellae	Curling of secondary lamellae	Epithelial lifting
Control	4 and 28 d	-	-	-	-
Acute (48 mg/L)	4 d-ZnO NPs	+	++	+	++
	4 d-ZnSO ₄	++	++	++	++
Sub-acute (4.8 mg/L)	28 d- ZnO NPs	+++	+++	+++	+++
-	28 d-ZnSO ₄	+++	+++	+++	+++
Depuration scenario					
Control	4 and 28 d	-	-	-	-
Acute	4 d-ZnO NPs	-	+	-	+
	4 d-ZnSO ₄	+	++	-	+
Sub-acute	28 d- ZnO NPs	++	+	+	+
	28 d-ZnSO ₄	++	++	+	++

Score value: None (-), mild (+), moderate (++) and severe (+++).

Table 2. Summary of histopathological effects in liver of *R. rutilus caspicus* after acute (48 mg/L, 4 d) and sub-acute (4.8 mg/L, 28 d) exposure to ZnO NPs and ZnSO₄ (Scenario B) and subsequent depuration (14 d) (Scenario C).

		Lesions			
Exposure scenario	Duration	Nuclear conjunction	Hypertrophy of hepatocytes	Hepatic lipolysis	Focal necrosis
Control	4 and 28 d	-	-	-	-
Acute (48 mg/L)	4 d-ZnO NPs	-	+	+	-
	4 d-ZnSO ₄	+	+	-	+
Sub-acute (4.8 mg/L)	28 d- ZnO NPs	+	++	+	++
	28 d-ZnSO ₄	++	++	++	+
Depuration scenario	_				
Control	4 and 28 d	-	-	-	-
Acute	4 d-ZnO NPs	-	-	-	-
	4 d-ZnSO ₄	-	+	-	-
Sub-acute	28 d- ZnO NPs	+	+	-	-
	28 d-ZnSO ₄	+	+	+	+

Table 3. Summary of histopathological effects in kidney of <i>R. rutilus caspicus</i> after acute (48
mg/L, 4 d) and sub-acute (4.8 mg/L, 28 d) exposure to ZnO NPs and ZnSO ₄ (Scenario B) and
subsequent depuration (14 d) (Scenario C).

		Lesions			
Exposure	Duration	Degeneration of	Degeneration of	Degeneration of	
scenario		Bowman's capsule	glomerulus	renal tubule	
Control	4 and 28 d	-	-	-	
Acute (48 mg/L)	4 d-ZnO NPs	+	+	-	
	4 d-ZnSO ₄	+	+	+	
Sub-acute (4.8 mg/L)	28 d- ZnO NPs	++	+++	-	
	28 d-ZnSO ₄	++	++	++	
Depuration scenario					
Control	4 and 28 d	-	-	-	
Acute	4 d-ZnO NPs	-	+	-	
	4 d-ZnSO ₄	-	-	+	
Sub-acute	28 d- ZnO NPs	-	+	-	
	28 d-ZnSO ₄	+	-	+	

Score value: None (-), mild (+), moderate (++) and severe (+++)

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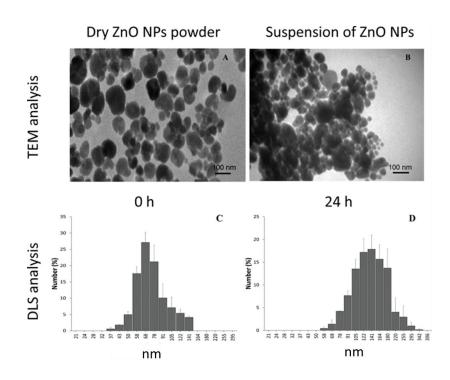


Figure 1. A) ZnO NPs dry powder; B) ZnO NPs in a water sample collected from fish tank (with fishes); C and D) hydrodynamic radius of ZnO NPs in a water sample collected from fish tank after 0 h (C) and 24 h (D) from dosing (48 mg/L ZnO NPs).

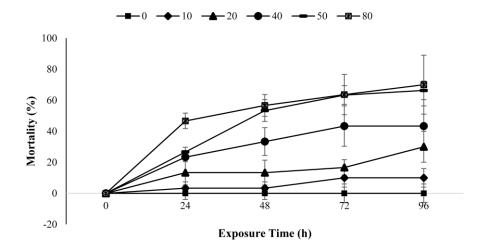


Figure 2. Mortality (%) in *R. rutilus caspicus* exposed to increasing ZnO NPs concentrations (0, 10, 20, 40, 50 and 80 mg/L ZnO NPs) (Scenario A).

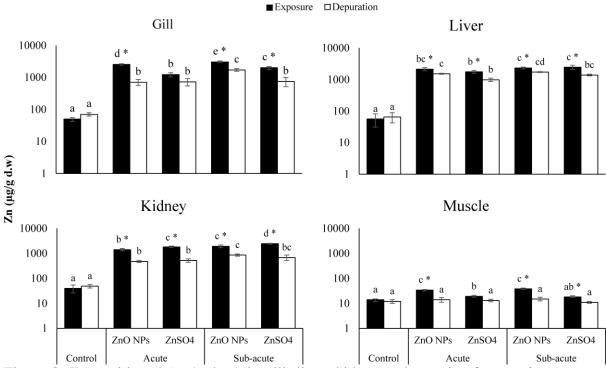


Figure 3. Zn partitioned (μ g/g d.w.) in gill, liver, kidney and muscle of *R. rutilus caspicus* after acute (48 mg/L, 4 d) and sub-acute (4.8 mg/L, 28 d) exposure to ZnO NPs and ZnSO₄ (Scenario B) and subsequent depuration (14 d) (Scenario C). Letters indicate statistically significant differences between treatments in each scenario (p < 0.05); asterisk (*) indicates statistically significant differences between scenario B and C for both ZnO NPs and ZnSO₄ at each exposure concentrations (p < 0.05).

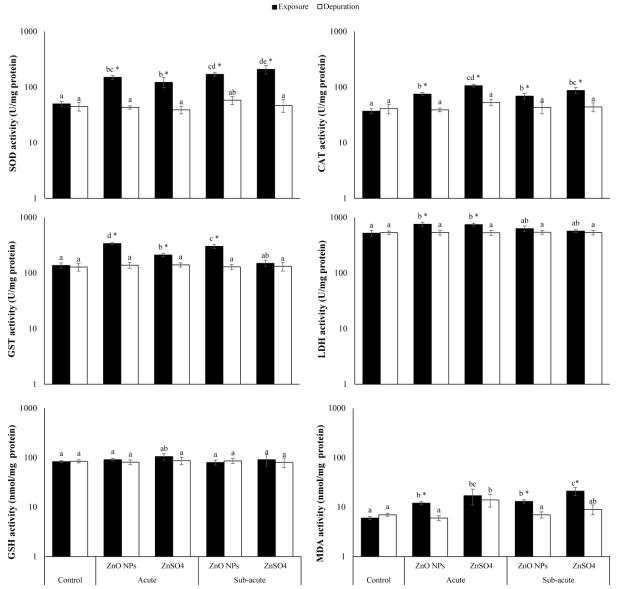


Figure 4. Biochemical parameters in liver of *R. rutilus caspicus* after acute (48 mg/L, 4 d) and sub-acute (4.8 mg/L, 28 d) exposure to ZnO NPs and ZnSO₄ (Scenario B) and subsequent depuration (14 d) (Scenario C). Letters indicate statistically significant differences between treatments in each scenario (p < 0.05); asterisk (*) indicates statistically significant differences between scenario B and C for both ZnO NPs and ZnSO₄ at each exposure concentration (p < 0.05) within the same target organ; U = 1 µmol/min.

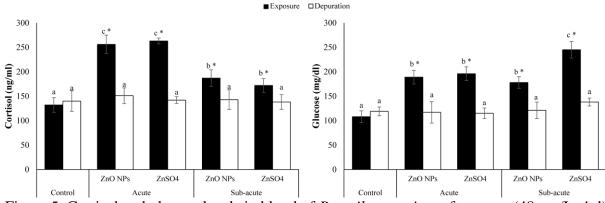


Figure 5. Cortisol and glucose levels in blood of *R. rutilus caspicus* after acute (48 mg/L, 4 d) and sub-acute (4.8 mg/L, 28 d) exposure to ZnO NPs and ZnSO₄ (Scenario B) and subsequent depuration (14 d) (Scenario C). Letters indicate statistically significant differences between treatments in each scenarios (p < 0.05); asterisk (*) indicates statistically significant differences between scenario B and C for both ZnO NPs and ZnSO₄ at each exposure concentration (p < 0.05).

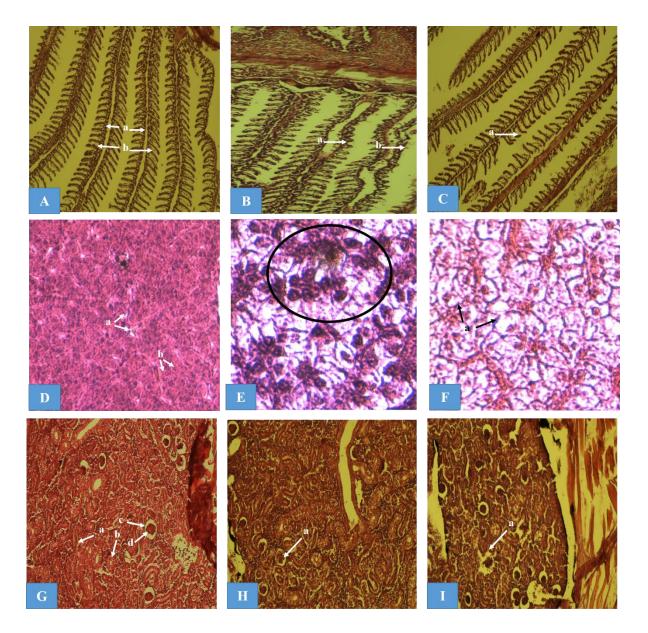


Figure 6. Histopathology of *R. rutilus caspicus* gill (A-C), liver (D-F) and kidney (G-I) tissues after hematoxylin and eosin staining considering acute or sub-acute exposures after depuration period. (A) control fish: (a) primary lamella, (b) secondary lamellae (100x); (B) sub-acute exposure to ZnO NPs (4.8 mg/L; 28 d): (a) shortening of secondary lamellae, (b) collapse of secondary lamellae (100x); (C) after depuration of sub-acute exposure (14 d): (a) curling of secondary lamellae (100x); (D) control fish: (a) sinusoidal portal blood, (b) hepatocytes (400x); (E) acute exposure to ZnO NPs (48 mg/L; 4 d): (circles) nuclear conjunction (1000x); (F) after depuration of sub-acute exposure (14 d): (a) hypertrophy of hepatocytes (1000x); (G) control fish: (a) longitudinal tubule, (b) tubule, (c) Bowman's capsule, (d) glomerulus (400x); (H) acute exposure to ZnO NPs (48 mg/L; 4 d): (a) degeneration of glomerulus (400x); and (I) sub-acute exposure to ZnO NPs (4.8 mg/L; 28 d): (a) degeneration of Bowman's capsule (400x).

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