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4 1 **Effect of Pulsed Electric Fields and High Pressure Homogenization on the**  
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6 2 **aqueous extraction of intracellular compounds from the microalgae *Chlorella***  
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8 3 ***vulgaris***  
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13 5 Daniele Carullo<sup>a</sup>, Biresaw Demelash Abera<sup>a</sup>, Alessandro Alberto Casazza<sup>b</sup>, Francesco Donsì<sup>a</sup>,  
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15 6 Patrizia Perego<sup>b</sup>, Giovanna Ferrari<sup>a,c</sup>, Gianpiero Pataro<sup>a\*</sup>  
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19 8 <sup>a</sup>Department of Industrial Engineering, University of Salerno, via Giovanni Paolo II, 132, 84084  
20  
21 9 Fisciano (SA), Italy  
22

23 10 <sup>b</sup>Department of Civil, Chemical and Environmental Engineering, University of Genoa, Via Opera  
24  
25 11 Pia 15, 16145, Genoa, Italy  
26

27 12 <sup>c</sup>ProdAl Scarl – University of Salerno, via Ponte don Melillo, 84084 Fisciano (SA), Italy  
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62 **14 Abstract**  
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64 15 Pulsed Electric Fields (PEF) and High Pressure Homogenization (HPH) are promising and scalable  
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66 16 cell disruption technologies of microalgae cells. In this work, the permeabilization degree,  
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68 17 morphological properties, and extractability of intracellular compounds from microalgae *Chlorella*  
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70 18 *vulgaris* suspensions (1.2%, w/w) were investigated as a function of PEF treatment at different  
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72 19 electric field strengths (10–30 kV/cm) and total specific energy input (20–100 kJ/kg), in comparison  
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74 20 with the more disruptive HPH treatment (150 MPa) at different number of passes ( $n_p=1-10$ ). The  
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76 21 conductivity and the particle size analyses, as well as the SEM images, clearly showed that PEF  
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78 22 induces the permeabilization of the cell membranes in an intensity-dependent manner, without  
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80 23 producing any cell debris, whereas HPH treatment causes the total disruption of the algae cells into  
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82 24 small fragments. Coherently with the lower permeabilization capability, PEF promoted the selective  
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84 25 extraction of carbohydrates (36 %, w/w, of total carbohydrates), and low molecular weight proteins  
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86 26 (5.2 %, w/w, of total proteins) with a relatively low energy input (2.9 kWh/kg<sub>DW</sub>), while HPH  
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88 27 required a significantly higher energy (20.0 kWh/kg<sub>DW</sub>) to induce the undifferentiated release of all  
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90 28 the intracellular content, resulting in a 1.1 and 10.3 fold higher yields than PEF, respectively of  
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92 29 carbohydrates and protein. These results suggest that, in a multi-stage biorefinery, PEF could  
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94 30 represent an energy-efficient cell disruption method for the selective recovery of small-sized  
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96 31 cytoplasmic compounds, while HPH should be placed at the end the cascade of operations allowing  
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98 32 the recovery of high molecular weight intracellular components.  
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105 34 **Keywords**— Microalgae; Pulsed electric fields; High pressure homogenization; Cell disruption;  
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107 35 **Proteins; Carbohydrates.**  
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115 39 **1. Introduction**  
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121 40 *Chlorella vulgaris* is a freshwater eukaryotic microalga with a mean diameter ranging from 2.5 to 5  
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123 41  $\mu\text{m}$  [1] belonging to the division of Chlorophyta. It has drawn large attention over the last decades  
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125 42 because of its capability to accumulate large amounts of valuable components, especially proteins (51  
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127 – 58 %), but also polyunsaturated fatty acids (14 – 22 %), carbohydrates (12 – 17 %), nucleic acids  
128 43  
129 (4 – 5 %), vitamins and minerals [2, 3]. Moreover, it accumulates also chlorophyll (1-2%) that imparts  
130 44  
131 the characteristic green color, masking the other less concentrated pigments, such as lutein and other  
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133 carotenoids [4]. The extraction of all these intracellular compounds, which can be used as natural  
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135 additives or active ingredients for food, cosmetic, pharmaceutical and animal feed products, as well  
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137 as in the production of biofuels [5, 6], is crucial for achieving an economically feasible microalgae  
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139 biorefinery [7].  
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143 50 However, these compounds are located in different parts of the cells, protected by the rigid cell wall  
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145 51 and membranes surrounding the cytoplasm and the internal organelles (e.g., chloroplast), which  
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147 52 greatly limit their rate of mass transfer during extraction. The more conventional extraction processes  
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149 53 from biomass, generally conducted on dry material, suffer from several limitations, namely the long  
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151 54 extraction times and the use of relatively large amounts of solvent, and usually lead to the co-  
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153 55 extraction of undesirable components, with increased downstream processing costs [7, 8]. In  
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155 56 particular, the drying of microalgal biomass is reported to be one of the major energy-consuming  
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157 57 steps within the overall process and is responsible for significant losses of valuable compounds [5,  
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162 59 For these reasons, the application of conventional or innovative cell disruption methods to wet  
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164 60 biomass may considerably promote the implementation of the biorefinery concept on microalgae,  
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166 61 enabling a faster and more efficient release of intracellular compounds at low temperature. This also  
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168 62 contributes to limit the degradation of the extracts and promotes the reduction of energy costs, of  
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170 63 solvent consumption, as well as of the extraction time [7, 8].

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172 64 Among the innovative cell disruption methods, the Pulsed Electric Fields (PEF) and the High Pressure  
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174 65 Homogenization (HPH) treatments have emerged as promising methods for the mild and complete

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180 66 disruption of biological cells, respectively [8, 9 – 13]. Moreover, both PEF and HPH enable the  
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182 67 treatment of wet biomass, avoiding the need for energy-intensive drying, and can be easily scaled up  
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184 68 to process large volumes [5, 14 – 16].

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187 69 In PEF processing, the biomaterial is placed between two electrodes of a treatment chamber and  
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189 70 exposed to high intensity electric fields (10-50 kV/cm), applied in the form of repetitive pulses of  
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191 71 very short duration (from several nanoseconds to few milliseconds), which induce the  
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193 72 permeabilization of cell membranes by electroporation, facilitating the subsequent release of  
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195 73 intracellular matter [17]. Several studies highlighted the effectiveness of PEF to enhance the selective  
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197 74 recovery of intracellular compounds from wet microalgal biomass, including lipids [18, 19], pigments  
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199 75 [8, 12, 20 – 22], carbohydrates, and water-soluble proteins of small molecular weight [6, 12, 13, 22,  
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201 76 23].

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204 77 However, the extraction of molecules of higher molecular weight, or more bounded to the  
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206 78 intracellular structure (e.g., proteins), requires the application of more effective cell disruption  
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208 79 techniques, such as HPH [8].

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210 80 HPH is a purely mechanical process, during which a liquid dispersion of plant material or a cell  
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212 81 biosuspension is forced by high pressure (50-300 MPa) through a micrometric disruption chamber,  
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214 82 where lamination occurs. As a result, the biological cell suspension is subjected to extremely intense  
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216 83 fluid-mechanical stresses (shear, elongation, turbulence, and cavitation), which cause the physical  
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218 84 disruption of the cell wall and membranes [15, 24, 25].

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221 85 Due to its high cell disruption efficiency [7], HPH is reported to markedly increase the extraction  
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223 86 yield of several intracellular compounds from microalgae [7, 12, 26 – 28]. However, the HPH  
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225 87 treatment causes the non-selective release of intracellular compounds, with the concurrent dispersion  
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227 88 of cell debris, complicating the downstream separation processes [12]. Moreover, because of the  
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229 89 intense interfacial shear stresses and inherent heating occurring in the homogenization valve [29 -  
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231 90 30], HPH also may cause the degradation of heat sensitive extracts (proteins, lipids, phenolics).

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91 Although several studies already highlighted the potential of PEF and HPH pre-treatments in the  
92 microalgae biorefinery, to date, only the study of Safi et al. [28] has addressed the comparison of their  
93 efficiency in terms of cell disintegration, energy input and release of soluble proteins from microalgae  
94 *Nannochloropsis gaditana*. However, suspensions of this microalgae were prepared from a frozen  
95 paste and at different biomass concentration for PEF (15-60 g<sub>DW</sub>/L) and HPH (100 g<sub>DW</sub>/L) treatments.  
96 Moreover, a deeper knowledge regarding the impact of these novel technologies at micro and macro  
97 scale is required, which is thoroughly necessary in view of their use in a cascade biorefinery approach  
98 of microalgae, where the control of the degree of cell breakage could be exploited to enable the fine  
99 tuning of the recovery process of intracellular components [6, 7, 31].  
100 Therefore, the aim of this study is to investigate comparatively the effects of the main process  
101 parameters of both PEF and HPH treatments on the cell disintegration degree, the energy  
102 consumption, and the release of intracellular compounds (ionic substances, proteins, and  
103 carbohydrates) from fresh *C. vulgaris*, in order to select, for each investigated technology, the best  
104 treatment conditions in the perspective of their implementation in a biorefinery scheme.

## 106 2. Materials and Methods

### 107 2.1. Microalgae and cultivation

108 The microalgal strain used in this study was *Chlorella vulgaris* (CCAP 211), purchased from the  
109 Culture Collection of Algae and Protozoa (Argyll, UK). It was cultivated in modified Bold's basal  
110 medium [32] at pH 7.0 ± 0.5, in a 5 L horizontal tubular photobioreactor illuminated by four 40 W  
111 fluorescent lamps from one side [33]. The composition (per liter of distilled water) of the modified  
112 medium was as follows: 1.5 g NaNO<sub>3</sub>, 0.45 g MgSO<sub>4</sub>·7H<sub>2</sub>O, 0.15 g NaCl, 0.45 g K<sub>2</sub>HPO<sub>4</sub>·3H<sub>2</sub>O,  
113 1.05 g KH<sub>2</sub>PO<sub>4</sub>, 0.15 g CaCl<sub>2</sub>·2H<sub>2</sub>O, 0.003 g vitamin B<sub>1</sub>, 7.5 10<sup>-6</sup> g vitamin B<sub>8</sub>, 7.5 10<sup>-6</sup> g vitamin B<sub>12</sub>  
114 and 6 mL of P-IV solution (Sigma Aldrich, Milan, Italy). The culture was aerated at a rate of 1000  
115 cm<sup>3</sup>/min with an air flow containing 2 % (v/v) carbon dioxide. Growth conditions were monitored by  
116 optical density (OD) measurements at 625 nm using a UV-Vis spectrophotometer (Lambda 25 model,

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298 117 Perkin Elmer, Milan, Italy). The pH of the culture medium was monitored during the experiments  
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300 118 using a pH meter (pH211, HANNA Instruments, Woonsocket, RI). Microalgae were harvested during  
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302 the end of the exponential phase at a biomass concentration of about 3 g<sub>DW</sub>/L of suspension and then  
303 119 concentrated by centrifugation (centrifuge model 42426, ALC, Milan, Italy) at 4000×g for 10 min at  
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305 120 20 °C up to a final concentration of 12 g<sub>DW</sub>/L. The concentrated biomass was pre-packed in high-  
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307 121 density polyethylene bottles (Nalgene) cooled at 4 °C, and sent to the laboratories of ProdAl Scarl  
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309 122 (University of Salerno, Italy). Samples were transported in an EPS box under refrigerated conditions  
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311 123 and delivered within 24 hours. PEF and HPH treatments were performed on the delivery day. The  
312  
313 124 initial electrical conductivity of algae suspension was about 1.78±0.03 mS/cm at 25 °C (Conductivity  
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315 125 meter HI 9033, Hanna Instrument, Milan, Italy).  
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## 321 322 128 **2.2. PEF Treatment**

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324 129 PEF treatments were conducted in a bench-scale continuous flow PEF unit, described in detail in a  
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326 130 previous work [6]. Briefly, the unit consisted of a peristaltic pump to control the flow rate of the algae  
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328 131 suspension through the system. The inlet temperature of the algae suspension was controlled using a  
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330 132 stainless steel coil immersed in a water heating bath. The PEF treatment zone consisted of two  
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332 133 modules, each made of two co-linear cylindrical treatment chambers, hydraulically connected in  
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334 134 series, with an inner radius of 1.5 mm and a gap distance of 4 mm. The treatment chambers were  
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336 135 connected to the output of a high voltage pulsed power (20 kV-100 A) generator (Diversified  
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338 Technology Inc., Bedford, WA, USA) able to deliver monopolar square pulses (1-10 μs, 1-1000 Hz).  
339 136 The maximum electric field intensity (E, in kV/cm) and total specific energy input (W<sub>T</sub>, in kJ/kg<sub>susp</sub>)  
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341 137 were measured and calculated as reported in Postma et al. [6]. T-thermocouples were used to measure  
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343 138 the product temperature at the inlet and outlet of each module of the PEF chamber.  
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347 140 During PEF treatment, the algae suspension (12 g<sub>DW</sub>/L) was pumped, from a feeding tank under  
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349 141 stirring, through the treatment chamber at a constant flow-rate of 33.3 mL/min. The pulse length was  
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351 142 fixed at 5 μs, while the electric field strength (E) of 10, 20 and 30 kV/cm and total specific energy

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357 143 input ( $W_T$ ) of 20, 60, and 100 kJ/kg<sub>g<sub>susp</sub></sub> were set by varying the applied voltage and the pulse repetition  
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359 144 frequency, respectively. All the experiments were carried out at an inlet temperature of each module  
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361 145 of PEF chamber of 25 °C, while the maximum temperature increase at the exit of each module due  
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364 146 to Joule effect never exceeded 10 °C.

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366 147 At the exit of the treatment chamber, treated and untreated (without applying PEF treatment) algae  
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368 148 suspensions were collected in plastic tubes and placed in an ice water bath to be rapidly cooled up to  
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370 149 a final temperature of 25 °C before undergoing to the aqueous extraction process.

### 372 150 373 374 151 **2.3. HPH treatment**

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376 152 HPH treatments were carried out by using an in-house developed laboratory scale high-pressure  
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379 153 homogenizer [34]. The *C. vulgaris* suspensions, at the same concentration as for PEF treatment tests  
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381 154 (12 g<sub>DW</sub>/L), were forced to pass through a 100 µm diameter orifice valve (model WS1973, Maximator  
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383 155 JET GmbH, Schweinfurt, Germany) upon pressurization by means of an air driven Haskel pump  
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385 156 (model DXHF-683, EGAR S.r.l., Milan, Italy). The pressure drop across the orifice and the  
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387 157 volumetric flow rate of the suspension were 150 MPa and 155 mL/min, respectively. In this work,  
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389 158 the algae suspensions were treated with a different number of passes ( $n_p = 1 - 10$ ). In order to prevent  
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391 159 excessive heating, after each pass, the suspensions were cooled at 25°C by passing through a tube-in-  
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393 160 tube heat exchanger, located downstream of the orifice valve.

### 394 160 395 396 161 397 398 162 **2.4. Water extraction**

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400 163 After processing, untreated and treated (PEF, HPH) samples were allowed to stand for 1 h at 25 °C  
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402 164 under shaking at 160 rpm to allow intracellular components to diffuse out of the cells. After this  
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404 165 resting time, the cell suspensions were centrifuged (10 min, 5700×g) (PK121R model, ALC  
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406 166 International, Cologno Monzese, IT) and the supernatant was transferred to fresh tubes and stored at  
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408 167 -20 °C until further analysis.

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416 **2.4.1. Electrical conductivity measurement**  
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418 Changing of the electrical conductivity ( $\sigma$ ) of untreated and treated (PEF, HPH) algae suspensions  
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420 was monitored periodically (Conductivity meter HI 9033, Hanna Instrument, Milan, Italy) over time  
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422 for up to 24 h by maintaining the samples in a water bath set at a constant temperature of 25 °C.  
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424 The collected data were used also to evaluate (Eq. 1) the cell disintegration index ( $Z_p$ ), which has  
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426 been successfully used as a reliable macroscopic indicator of the degree of cell membrane  
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428 permeabilization induced by PEF [35-36]:  
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$$Z_p = \frac{\sigma_{PEF,t} - \sigma_0}{\sigma_{MAX} - \sigma_0} \quad (1)$$
  
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434 where  $\sigma_{PEF,t}$  is the electrical conductivity of PEF treated biosuspensions measured at time t,  $\sigma_0$  is the  
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436 conductivity of untreated algae suspension at time 0, and  $\sigma_{MAX}$  is the conductivity of biosuspension  
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438 with completely disrupted algae cells (HPH treatment: P = 150 MPa,  $n_p = 5$ ). The Eq. (1) gives  $Z_p=0$   
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440 for intact algae cells and 1 for fully disrupted cells.  
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443 **2.4.2. Particle size distribution (PSD) analysis**  
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446 PSD of untreated and treated (PEF or HPH) algae suspensions were analyzed by laser diffraction at  
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448 25 °C, using a MasterSizer 2000 particle size analyzer (Malvern, United Kingdom). Using the  
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450 Fraunhofer approximation, which does not require the knowledge of the optical properties of the  
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452 sample, the size distribution of the algal suspension was determined, from which the mean particle  
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454 size expressed as volume moment mean diameter ( $D_{4,3}$ ) was evaluated for each processing condition.  
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456 The parameters used in the determination of the PSD were the properties of water at 25 °C (refraction  
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458 index = 1.33), which was used as dispersant medium.  
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462 **2.4.3. Scanning Electron Microscopy (SEM) analysis**  
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465 The morphological features and cellular details of algae cells were analyzed by using a Scanning  
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467 Electron Microscopy (SEM). Pellets derived from the centrifugation of untreated and treated (PEF or  
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475 194 HPH) algae suspensions were prepared as described by Kunrunmi et al. [37] with some modifications.  
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478 195 At first, samples were fixed by immersion in a 2 % (v/v) glutaraldehyde phosphate buffer solution.  
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480 196 The buffer was then removed and the pellets were osmotically dehydrated with ethanol solutions of  
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482 197 increasing concentration (25%, 50%, 75%, and 100% (v/v)). Afterwards, ethanol was removed from  
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484 198 the pellet with supercritical CO<sub>2</sub> in a Quorum K850 critical point dryer (Quorum Technologies Ltd,  
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486 199 London, UK) and the latter was then metallized by means of the Agar Auto Sputter Coater 103A  
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488 200 (Agar Scientific Ltd, Stansted, UK), before being analysed in a high-resolution ZEISS HD15  
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490 201 Scanning Electron Microscope (Zeiss, Oberkochen, Germany).

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#### 494 203 **2.4.4. Dry Matter (DM) content analysis**

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497 204 Approximately 40 mL of the supernatants collected from the centrifugation of untreated and treated  
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499 205 (PEF or HPH) algae suspension were placed in aluminum cups and dried in an oven (Heraeus,  
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501 206 Germany) at 80 °C until constant mass was achieved. DM was gravimetrically determined by  
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503 207 weighing the samples before and after drying on an analytical balance (Gibertini, Italy). The dry mass  
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505 208 content was expressed as g of dry matter/kg of supernatant (g<sub>DW</sub>/kg<sub>SUP</sub>).

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#### 509 210 **2.4.5. Proteins Analysis**

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511 211 The water-soluble protein concentration in the supernatants was evaluated using the Lowry method  
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513 212 [38], with some modifications. The Folin-Ciocalteu reactive [39], purchased from Sigma Aldrich  
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516 213 (Milan, Italy), was initially diluted in two volumes of ultra-pure water (1:2, v/v); then 0.5 mL of the  
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518 214 diluted reactive were added to 1 mL of supernatant, previously mixed with 5 mL of the reactive “C”  
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520 215 [50 volumes of reactive “A” [(2% (w/v) Na<sub>2</sub>CO<sub>3</sub> + 0,1 N NaOH) + 1 volume of reactive “B” (1/2  
521  
522 216 volume of 0.5% (w/v) CuSO<sub>4</sub> ·5H<sub>2</sub>O + 1/2 volume of 1% KNaC<sub>4</sub>H<sub>4</sub>O<sub>6</sub> ·4H<sub>2</sub>O)] (Sigma Aldrich,  
523  
524 217 Milan, Italy). Absorbance was measured at 750 nm against a blank (5 mL reactive “C” + 1 mL  
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526 218 deionized water + 0.5 mL Folin-Ciocalteu reactants) 35 min after the start of the chemical reaction  
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528 219 by using a V-650 Spectrophotometer (Jasco Inc. Easton, MD, USA). Bovine serum albumin (BSA)

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534 220 (A7030, Sigma Aldrich, Milan, Italy) was used as standard and the results were expressed as mg  
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536 221 equivalent of BSA per g<sub>DW</sub> .  
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#### 540 223 **2.4.6. Carbohydrates Analysis**

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543 224 The total carbohydrates concentrations of the supernatants were analyzed according to the method of  
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545 225 DuBois et al. [40]. 0.2 mL of 5% (w/w) phenol and 1 mL of concentrated sulfuric acid (Sigma Aldrich,  
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547 226 St. Louis, USA) was added to 0.2 mL of diluted supernatant (Dilution Factor = 5). Samples were  
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549 227 incubated at 35 °C for 30 min before reading the absorbance at 490 nm against a blank of 0.2 mL 5%  
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551 228 (w/w) phenol, 1 mL concentrated sulfuric acid and 0.2 mL of deionized water. D-Glucose (G8270,  
552  
553 229 Sigma-Aldrich, Milan, Italy) was used as a standard and the results were expressed as equivalent mg  
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555 230 of D-glucose per g of dry biomass.  
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#### 559 232 **2.5. Statistical Analysis**

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562 233 All treatments and analyses were performed in triplicate and the results were reported as mean values  
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564 234 with their respective standard deviations. Statistically significant differences ( $p \leq 0.05$ ) between the  
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566 235 means were evaluated using one-way analysis of variance (ANOVA), performed with SPSS 20  
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568 236 (SPSS Inc., Chicago, USA) statistical package, and the Tukey's test.  
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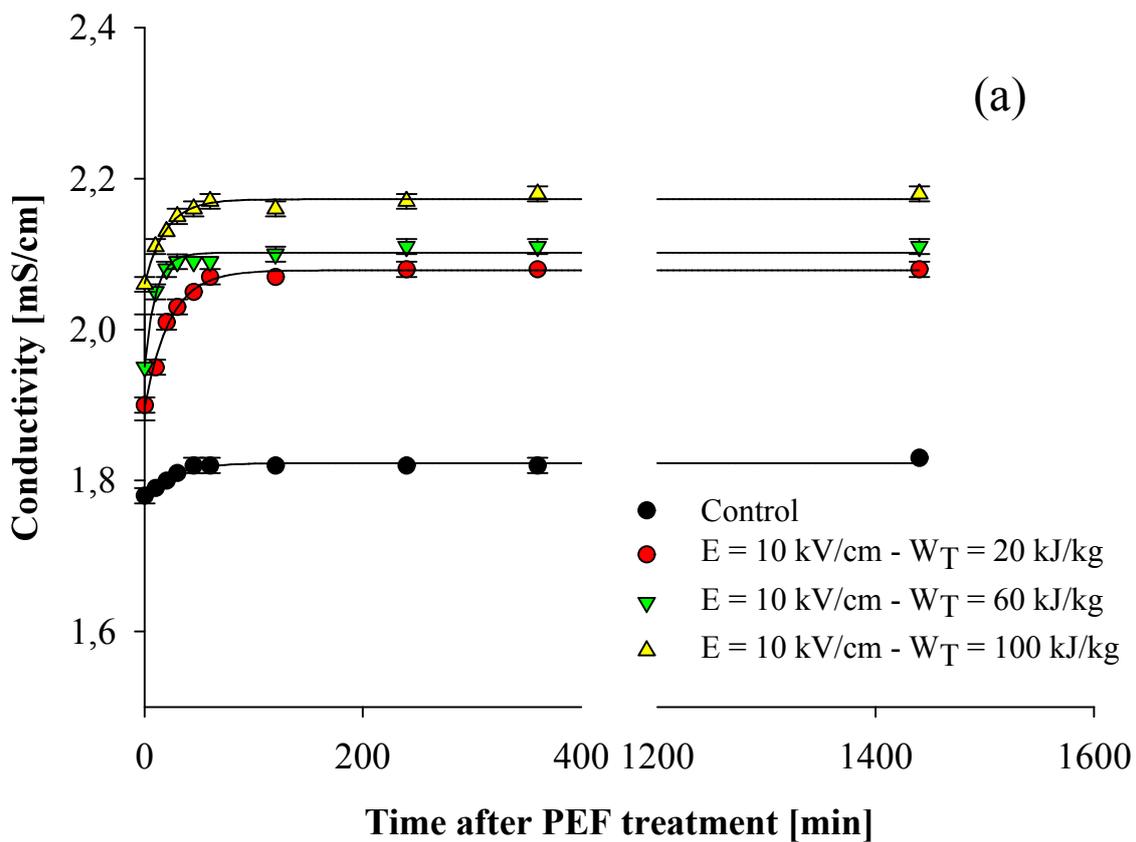
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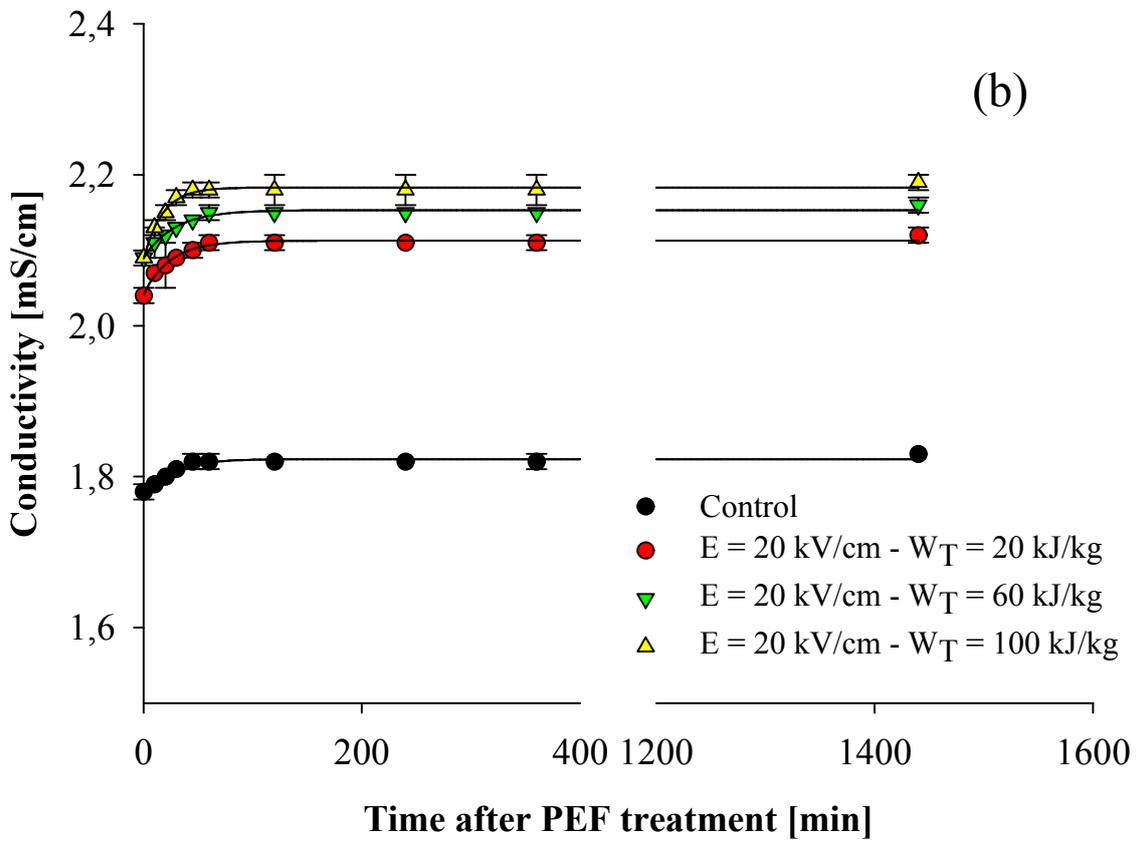
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593 **3. Results and Discussion**  
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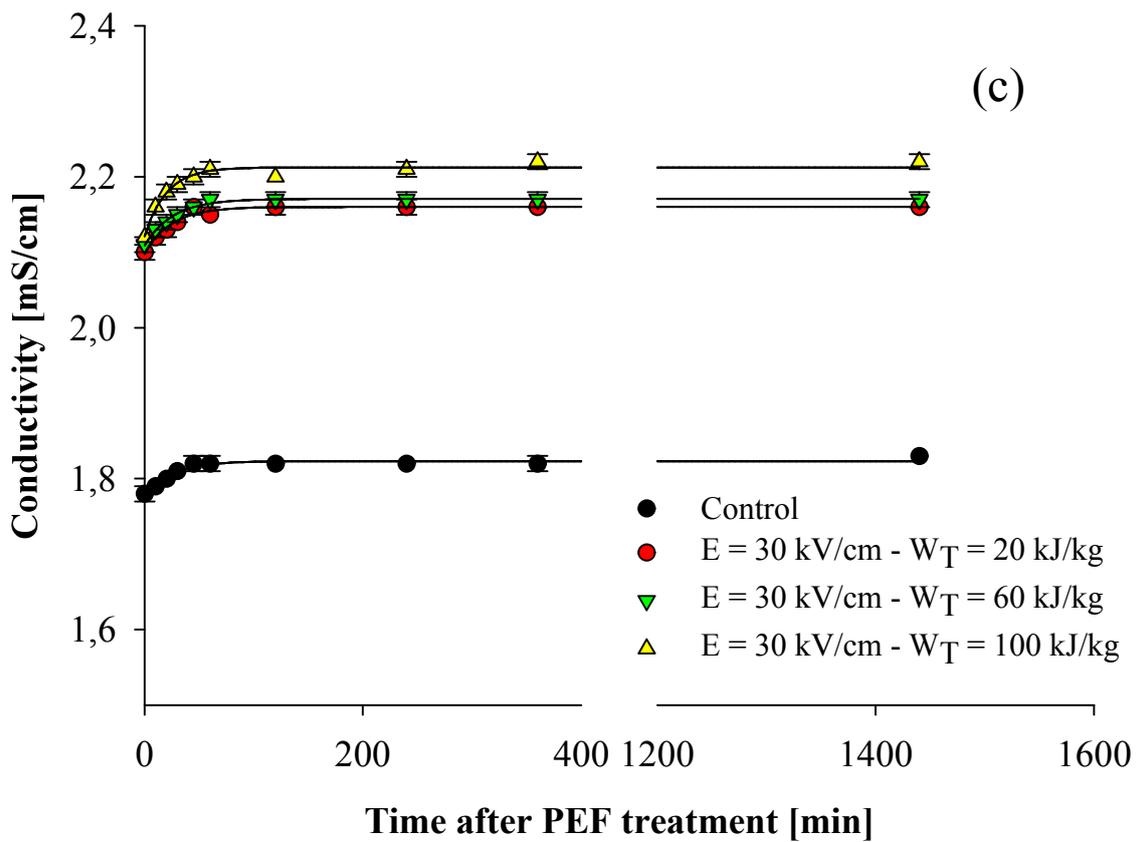
595 **3.1. Impact of PEF and HPH treatments on the release of ionic intracellular**  
596 **components**  
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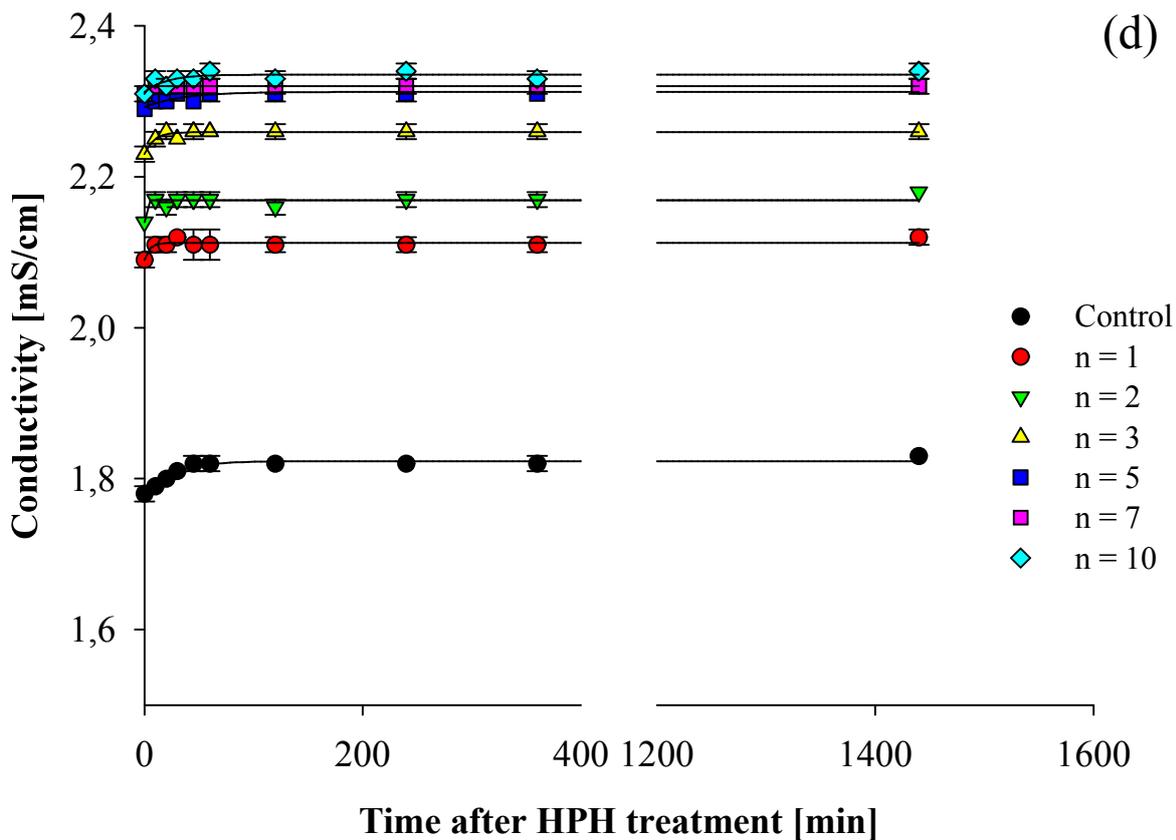
599  
600 241 The results of the measurements of the electrical conductivity of microalgae suspension have been  
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602 242 successfully used as a valuable indicator to assess and quantify the amount of ionic intracellular  
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604 243 components released from algae upon the application of the different cell disruption methods [12, 13,  
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606 244 23, 41].  
607

608 245 Figure 1 shows the effect of PEF treatment intensity ( $E$ ,  $W_T$ ), as well as the number of HPH passes  
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610 246 ( $n_p$ ) on the conductivity profiles of *C. vulgaris* suspensions over time at 25 °C.  
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**Fig.1.** Effect of incubation time after PEF and HPH treatment on electrical conductivity at 25 °C of (a-c) PEF ( $E=10-30$  kV/cm;  $W_T=20-100$  kJ/kg) and (d) HPH (150 MPa;  $n_p=1-10$ ) treated *C. vulgaris* suspension at a different number of passes. Control means untreated suspension.

For the sake of comparison, in the same graphs, also the time-conductivity profile of the untreated algae suspension is shown. Results demonstrate that the initial conductivity (1.78 mS/cm) of untreated suspension increased only slightly with the incubation time, likely due to a spontaneous release of a small fraction of intracellular ionic compounds, reaching a saturation value (1.82 mS/cm) already after 30 min of incubation.

The electroporation effect induced by the application of PEF treatment at different field strength (10-30 kV/cm) and energy input (20-100 kJ/kg) promoted a rapid release of the ionic intracellular compounds, which resulted in a substantial increase in the electrical conductivity, with respect to the untreated suspension (Figs. 1a-c). After PEF treatment, the saturation value was reached after 1 h of

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829 266 incubation. In this time interval, an increase of the field strength and energy input led to a faster  
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832 267 diffusion of the ionic intracellular substances into the aqueous phase, which is in agreement with the  
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834 268 electroporation theory. A further increase of the incubation time did not cause any significant increase  
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836 269 in the conductivity value, which leveled off to a final value in the range between 2.08- and 2.21  
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838 270 mS/cm, depending on the PEF treatment intensity applied.  
839  
840 271 A progressive increase of the content of ionic compounds in the extracellular medium upon PEF  
841  
842 272 treatment was also observed by Goettel et al. [23], which reported that the 79% of the total released  
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844 273 ions from *Auxenochlorella protothecoides* already occurred in the first hour after treatment. Similarly,  
845  
846 274 Postma et al. [6] and Pataro et al. [13] reported that PEF caused the permeabilization of the *C. vulgaris*  
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848  
849 275 cells and that an incubation time of 1 h was sufficient to allow small ions to diffuse out of the cells,  
850  
851 276 in agreement with the results reported in Figs. 1a-c.  
852  
853 277 The persistent continuous leakage of ionic compounds from the intracellular space during the first  
854  
855 278 hour after PEF treatment indicates the achievement of an irreversible electroporation [23]. However,  
856  
857 279 in addition to what reported to date by the literature on the topic, the results of Figs. 1a-c also suggest  
858  
859 280 that, although PEF markedly improved the mass transfer rate of ionic compounds, their diffusion  
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861 281 appears still hindered by the cell structure, which presumably is only partially damaged by the  
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863 282 electrical treatment, especially at the mildest treatment conditions.  
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865 283 Coherently with this assumption, when compared to PEF treatments, the HPH treatments resulted in  
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867  
868 284 a significant increase in the conductivity of *C. vulgaris* suspension, whose extent was greater when  
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870 285 increasing the number of HPH passes, as shown in Fig. 1d. More specifically, the mechanical  
871  
872 286 disruption of the algae cells appeared to be extremely fast, leading to an almost instantaneous  
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874 287 diffusion of the intracellular compounds into the aqueous phase, as observed also by Safi et al. [26].  
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876 288 Considering that HPH is a purely mechanical on-off disruption process [13], it is likely that after each  
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878 289 pass a certain fraction of algae cells are completely broken, while the residual cells remain intact, in  
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880 290 agreement with the observation of the significant extraction yield of ionic compounds after the multi-  
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882 291 pass HPH treatment, as reported in Fig. 1d.  
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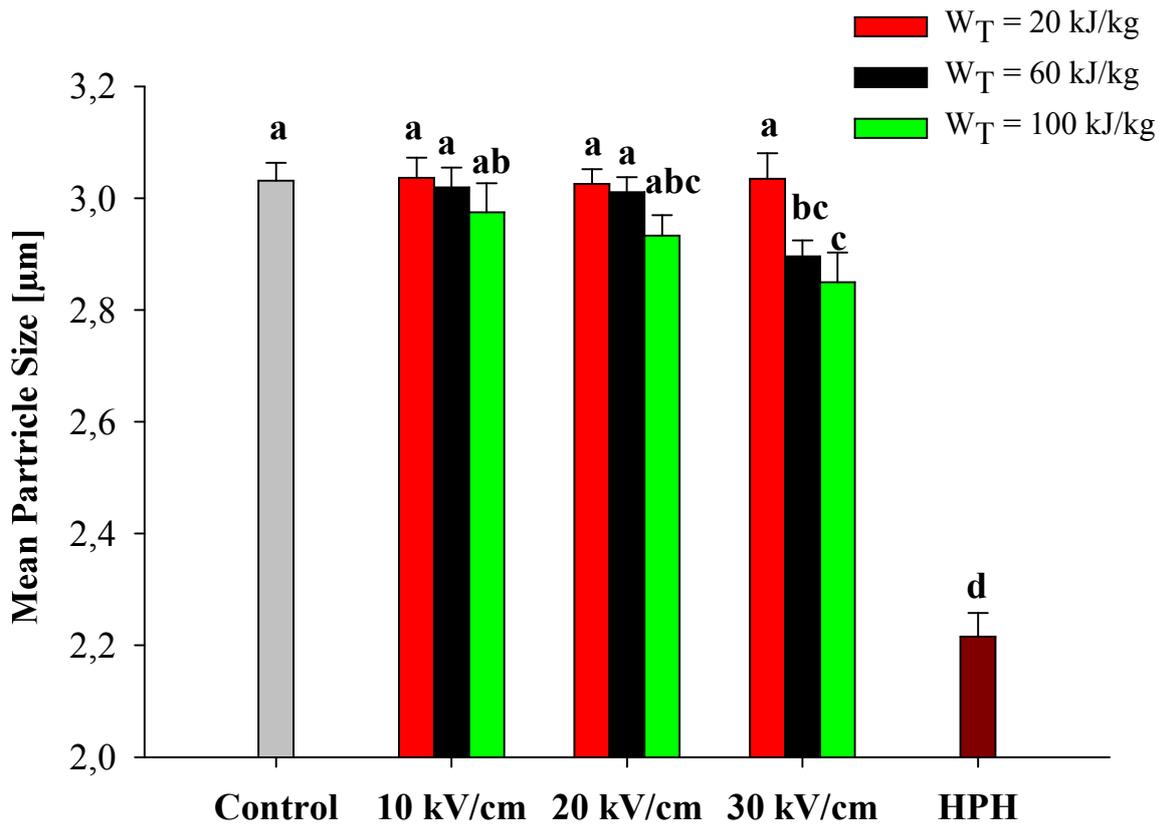
292 Coherently, the results of Fig. 1d also show that above 5 passes, the conductivity did not change  
293 significantly, and tended to an asymptotic value of 2.3 mS/cm, because the residual fraction of intact  
294 cells has become extremely small. However, such asymptotic value was significantly higher than that  
295 measured after the application of the most intense PEF treatment, confirming that the release of ionic  
296 compounds by PEF is incomplete.

297 Thus, setting the conditions of  $Z_p=1$  in correspondence of 5 HPH passes, the cell disintegration  
298 efficiency of PEF varied in a range dependent on the treatment intensity applied: the lowest value of  
299  $Z_p$  (0.47) was observed for a PEF treatment intensity of 10 kV/cm and 20 kJ/kg, whereas, increasing  
300 the electric field strength and energy input, a maximum  $Z_p$  value of 0.85 was recorded.

### 301 302 **3.2. Effect of PEF and HPH treatment on *C. vulgaris* cell structure**

303 In this work, particle size distribution (PSD) analyses and microscopic observations were carried out  
304 in order to gain insight on the impact of PEF and HPH treatments on the size and structure of *C.*  
305 *vulgaris* cells.

306 Fig. 2 depicts the mean particle size  $D_{4,3}$  for untreated (control), PEF treated at variable field strength  
307 and energy inputs, and HPH ( $n_p = 5$ ) treated microalgae suspensions.



**Fig. 2.** Mean particle size of untreated (control), PEF treated ( $E=10\text{-}30$  kV/cm;  $W_T=20\text{-}100$  kJ/kg) and HPH treated ( $P = 150$  MPa;  $n_p = 5$ ) *C. vulgaris* suspension. Different letters above the bars indicate significant differences between the mean values ( $p \leq 0.05$ ).

The PSD curves of untreated algae suspension revealed the presence of a single peak between 1 and 10  $\mu\text{m}$  (data not shown), which was characterized by a mean cell size of  $3.03 \pm 0.03$   $\mu\text{m}$  (Fig. 2).

The size distribution curve of PEF-treated algae suspension was very similar to that of the untreated sample (data not shown), showing only a slight decrease of the mean cell size with increasing the treatment intensity ( $E$  and  $W_T$ ). In fact, the value of the mean cell size significantly ( $p \leq 0.05$ ) decreased by about 6% only upon the application of the most intense PEF treatment conditions ( $E=30$  kV/cm,  $W_T \geq 60$  kJ/kg) (Fig. 2). These results seem to confirm that PEF is a relatively mild cell disruption method, preserving the initial structure of the algae cells.

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1006 321 The application of 5 HPH passes, instead, led to a significant change in the PSD curves of the  
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1008 322 microalgae suspension, highlighting a bimodal distribution, in which a second peak between 0.1 and  
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1011 323 1  $\mu\text{m}$  appeared (data not shown). As a result, a strong reduction in the mean cell size down to a value  
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1013 324 of  $2.22 \pm 0.04 \mu\text{m}$  was observed (Fig. 2), which is likely due to the complete cell disruption and the  
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1015 325 consequent formation of cell debris.

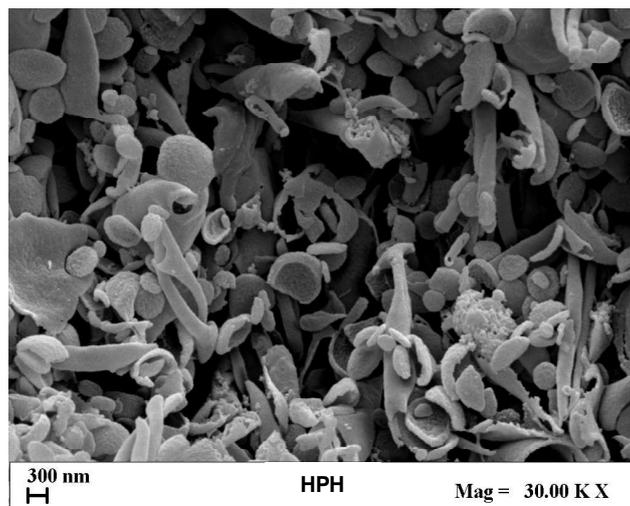
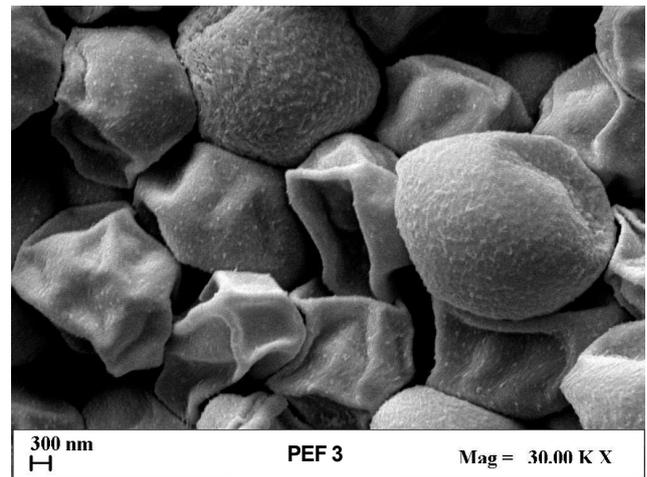
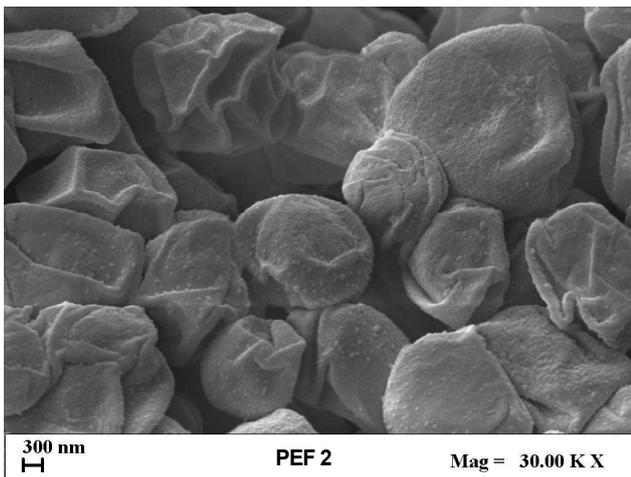
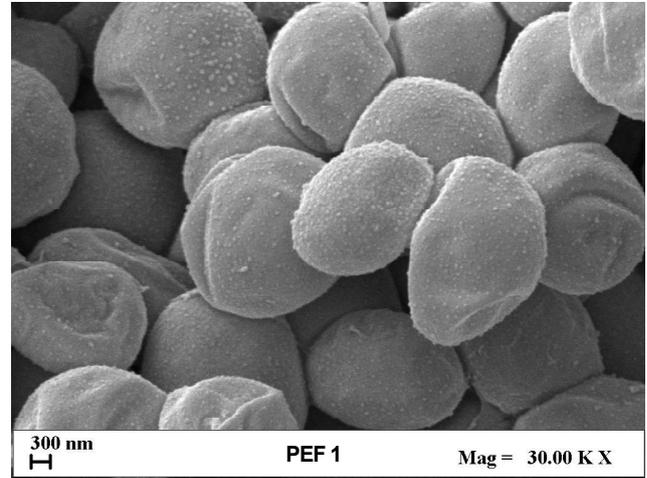
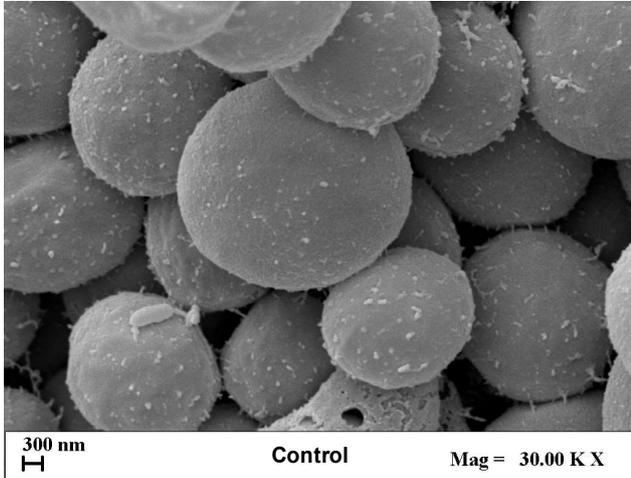
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1017 326 Partially in contrast with these results, Spiden et al. [42] found that the effect of an HPH treatment on  
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1019 327 *Chlorella* microalgae at different pressures ( $P = 30 - 107 \text{ MPa}$ ) only led to a slight decrease in the  
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1021 328 mean cell size, which was in agreement with the only partial fragmentation achieved. Eventually, in  
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1023  
1024 329 our case, the application of a higher pressure ( $P = 150 \text{ MPa}$ ) was capable of inducing the complete  
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1026 330 disruption of the cells, which is in agreement with the previous findings of Safi et al. [28]. Similarly,  
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1028 331 Shene et al. [27] and Samarasinghe et al. [16], studying the effect of HPH processing ( $P = 70 - 310$   
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1030 332  $\text{MPa}$ ,  $n_p = 1 - 6$ ) on *Nannochloropsis oceanica* microalgae, reported that the cells were fully disrupted  
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1032 333 in fragments, with a corresponding decrease in mean particle size.

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1034 334 In order to better interpret the results of Figs. 1 and 2, also SEM analyses were carried out on  
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1036 335 untreated, PEF-treated ( $E = 20 \text{ kV/cm}$ ;  $W_T = 20 - 100 \text{ kJ/kg}$ ), and HPH-treated ( $n_p=5$ ) microalgae,  
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1038 336 as shown in Fig. 3.

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1040 337 Untreated *C. vulgaris* cells exhibited their characteristic near-spherical shape and a diameter ranging  
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1042 338 from 1.5 and 4.5  $\mu\text{m}$ , which relate to the findings reported in the current literature [43].

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1044 339 The SEM images of Fig 3 clearly show the different impact of PEF and HPH treatments on the  
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1046 340 microalgal cell structure. Interestingly, the results clearly show, for the first time, the occurrence of a  
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1048 341 shrinkage phenomenon in PEF-treated algae cells, which, gradually lose their initial near-spherical  
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1050 342 shape with increasing the applied energy input but were never disintegrated into cell debris. The  
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1052 343 observed shrinkage could be associated with the partial release of the intracellular compounds through  
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1054 344 the electroporated cell membranes (Fig. 1b), which led in some cases to cell collapse (Fig. 2). Similar  
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1056 345 results were observed at different electric field strengths (data not shown).

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1066 346 In contrast, a complete disruption of the cells and the formation of small fragments was observed  
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1068 347 after 5 passes HPH treatment, which was consistent with the results of Figs. 1 and 2.  
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1124  
1125 **Fig.3.** Scanning electron microscopy (SEM) of *C. vulgaris* cells before (Control) and after PEF (20  
1126  
1127 351 kV/cm) at total specific energy input of 20 kJ/kg (PEF1), 60 kJ/kg (PEF2), 100 kJ/kg (PEF3), and  
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1129 352 HPH (P = 150 MPa;  $n_p = 5$ ) treatment of the microalgal suspension.

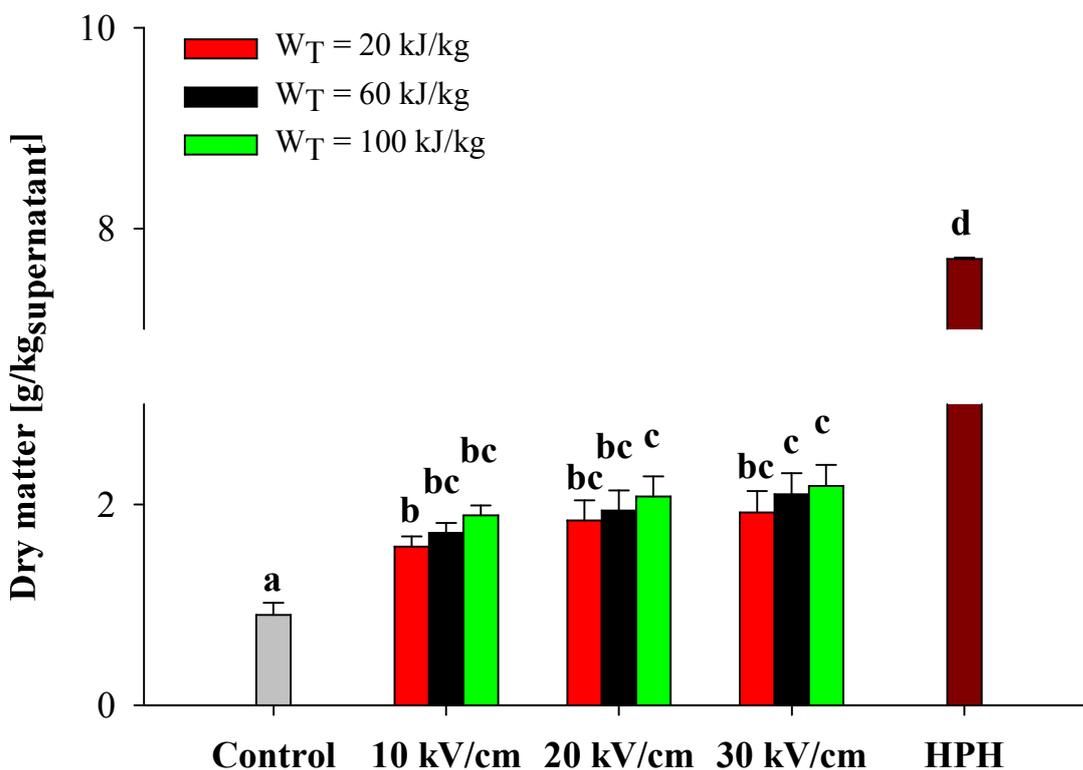
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1131 353  
1132  
1133 354 Similarly, the formation of cell fragments was observed by other authors upon the application of HPH  
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1135 355 treatments to *Chlorella* [26, 44] and *Neochloris abundans* [45] microalgae, highlighting the strong  
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1137 356 efficacy of HPH treatment as a method of complete cell disruption.

### 1138 1139 357 1140 1141 358 **3.3. Influence of PEF and HPH treatments on the release of intracellular compounds**

1142  
1143 359 The cell disruption efficiency of PEF and HPH treatments were also compared by monitoring the  
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1145 360 extractability of intracellular compounds by dry matter analyses and by measuring the amount of  
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1148 361 water-soluble compounds (proteins and carbohydrates) released into the supernatants obtained from  
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1150 362 untreated and treated (PEF, HPH) algae suspension.

#### 1151 1152 363 **3.3.1. Dry Matter of supernatants**

1153  
1154 364 The total amount of released intracellular compounds was evaluated by measuring the dry matter  
1155  
1156 365 content in the supernatant of untreated, PEF-treated at different field strength and energy inputs, and  
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1158 366 HPH-treated ( $n_p=5$ ) microalgae suspensions.



**Fig. 4.** Dry matter content in the supernatant of untreated (Control) and treated *C. vulgaris* suspension 1 h after PEF ( $E=10-30$  kV/cm;  $W_T=20-100$  kJ/kg) or after HPH ( $P = 150$  MPa;  $n_p = 5$ ) treatment. Different letters above the bars indicate significant differences between the mean values ( $p \leq 0.05$ ).

The results showed in Fig. 4 are in agreement with the conductivity measurements of Fig. 1. The application of PEF treatment markedly increased the dry matter content, when compared with the untreated sample. A higher field strength and energy inputs resulted in a higher degree of membrane permeabilization, leading to a significantly ( $p \leq 0.05$ ) higher release of intracellular compounds into the aqueous phase. The maximum value of dry matter content was detected at the most intense PEF treatment conditions ( $E = 30$  kV/cm;  $W_T = 100$  kJ/kg), which was 2.4 times higher than that detected in the supernatant of the untreated microalgae suspension. However, among PEF treated samples, statistically significant differences ( $p < 0.05$ ) were observed only between samples treated at 10 kV/cm and 20 kJ/kg with those treated either at 20 kV/cm and 100 kJ/kg or at 30 kV/cm for an energy

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1243 381 input greater than 20 kJ/kg. Remarkably, the results of Fig. 4 are in agreement with the previous  
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1245 382 findings of Goettel et al. [23]. The authors observed a continuous increase of cell components in the  
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1247 383 medium surrounding *Auxenochlorella protothecoides* when the energy input was increased up to 200  
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1249 384 kJ/kg at a constant field strength (34 kV/cm). Moreover, in our case, the release of intracellular  
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1251 385 soluble compounds by PEF varied in the range 13 – 18 % of total cell dry weight, which is also in  
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1253 386 agreement with the results obtained by Goettel et al. [23], who found that a PEF treatment at 30.5  
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1255 387 kV/cm and 155 kJ/kg caused the spontaneous release of intracellular matter up to 15% of the initial  
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1257 388 biomass dry weight (109 g/kg<sub>DW</sub>). Pataro et al. [13] also observed a slightly higher leakage of  
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1259 389 intracellular matter from *C. vulgaris* cells with increasing the field strength (from 27 to 35 kV/cm)  
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1261 390 and energy input (from 50 to 150 kJ/kg).

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1263 391 The stronger cell disintegration effect, achieved after 5 passes HPH treatment (Figs. 1-3), led to a  
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1265 392 highly efficient extraction of intracellular matter (Fig. 4), whose extent reached up to 64% of the total  
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1267 393 cell dry weight.

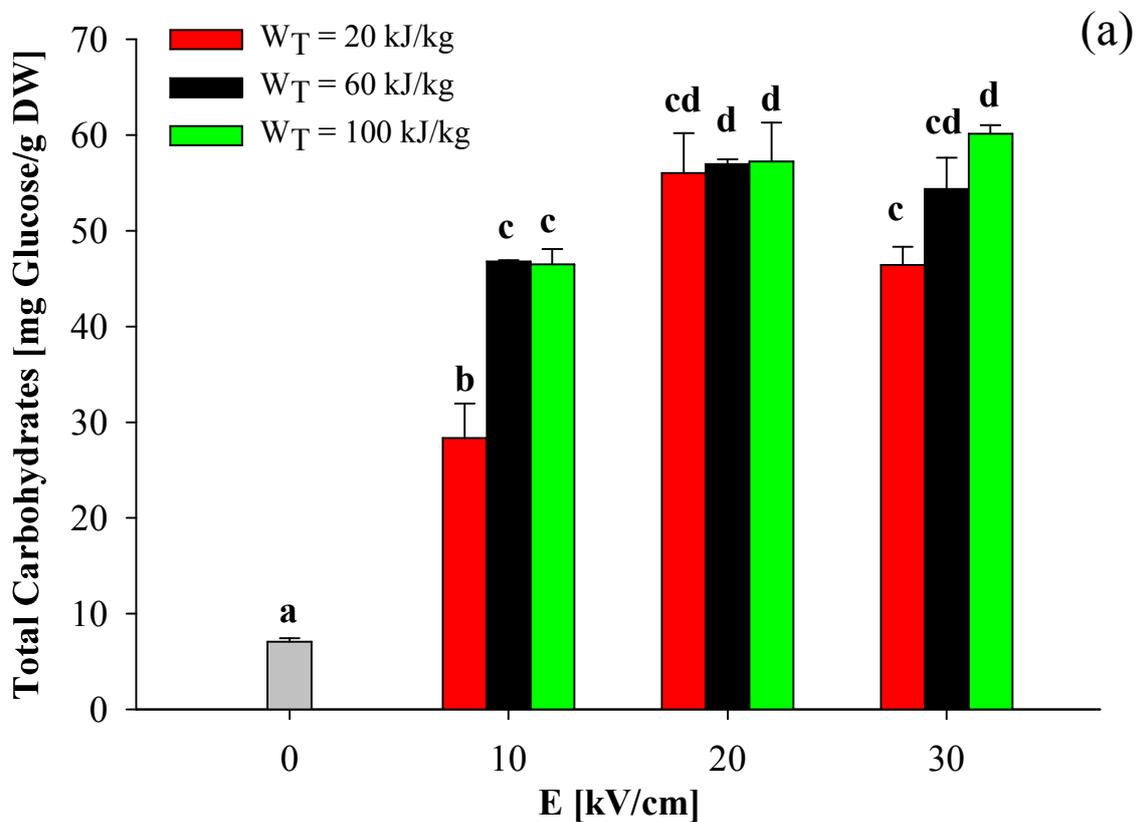
1268  
1269 394 The results of Fig. 4 were also confirmed by visual observation of the supernatants. In fact, while the  
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1271 395 supernatants obtained from centrifugation of fresh and PEF treated microalgal suspensions appeared  
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1273 396 colorless, those obtained from HPH treated samples were characterized by a green color (data not  
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1275 397 shown). This was likely due to the presence of cell debris (Fig. 3) containing green pigments, which,  
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1277 398 being extremely reduced in size, did not precipitate in the pellet after centrifugation [26]. With this  
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1279 399 assumption, it can be stated that part of the supernatant dry matter content from the HPH treated cells  
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1281 400 could be due to the presence of submicrometric residues, which remained suspended in the aqueous  
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1283 401 phase, making the downstream separation processes more difficult.

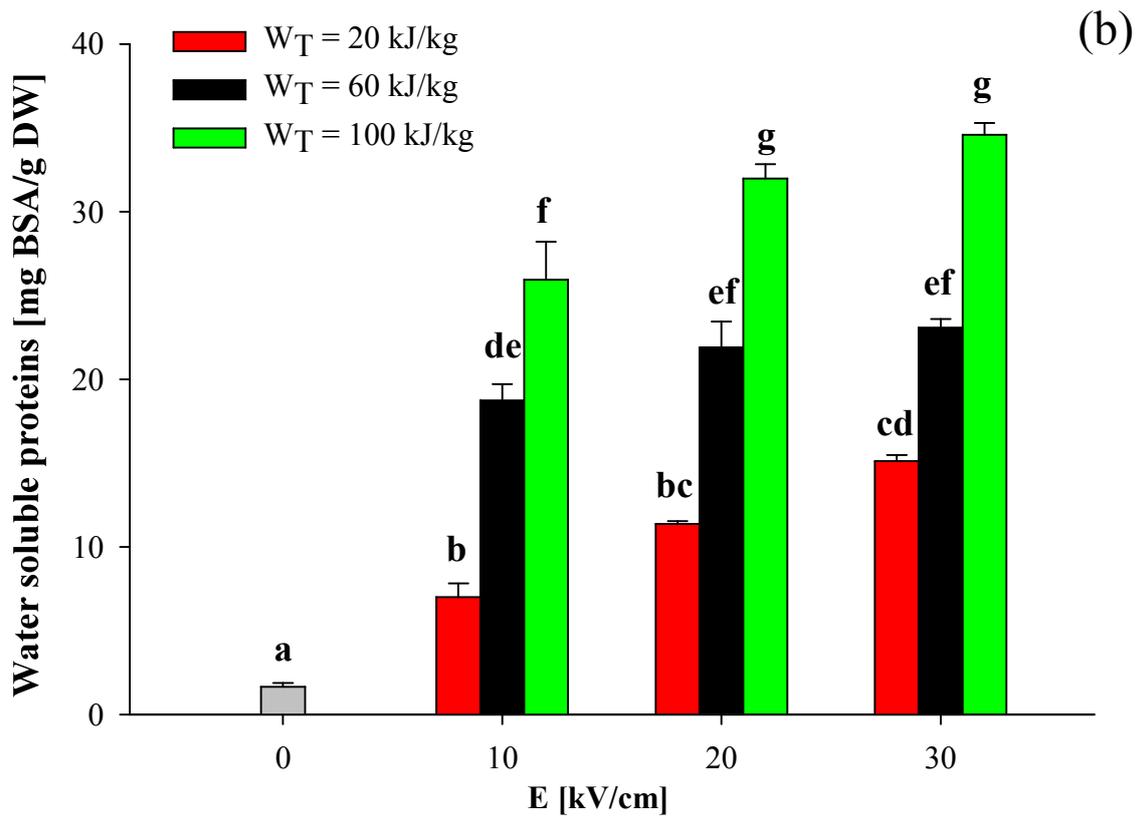
### 1284 1285 402 1286 1287 403 **3.3.2. Extractability of carbohydrates and proteins**

1288  
1289 404 Fig. 5 shows the concentration (on DW basis) of carbohydrates (Fig. 5a) and proteins (Fig. 5b)  
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1291 405 detected in the aqueous supernatant obtained 1 h after PEF treatment of *C. vulgaris* suspensions at  
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1293 406 different field strength and energy input.

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1302 407 When no PEF treatment was applied, only very low amounts of carbohydrates (7.06 mg/g<sub>DW</sub>) and  
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1304 408 proteins (1.65 mg/g<sub>DW</sub>) were released in the aqueous phase, which may be ascribed to either a  
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1306 409 concentration gradient across the intact cell membranes or to a spontaneous cell lysis.

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1308 410 The permeabilization effect of the cell membranes induced by the application of PEF treatment,  
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1310 411 instead, improved the mass transfer of intracellular compounds, leading to a significantly ( $p \leq 0.05$ )  
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1312 412 higher content of both carbohydrates and proteins, as compared to the untreated samples, being the  
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1314 413 extraction efficiency increased up to 20-fold for proteins and 8-fold for carbohydrates.  
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**Fig. 5.** Concentration of carbohydrates (a) and proteins (b) in the supernatant of untreated (0 kV/cm) and treated *C. vulgaris* suspension 1 h after PEF treatment as a function of the field strength and for different energy input. Different letters above the bars indicate significant differences between the mean values ( $p \leq 0.05$ ).

Among the PEF treated samples, the effect of the field strength applied (Fig. 5) appeared less important than that of the energy input within the investigated range, especially for the protein extraction, which is in agreement with previous findings [13, 41]. In particular, a significant ( $p \leq 0.05$ ) increase in the content of both intracellular compounds was detected only when the field strength was increased from 10 to 20 kV/cm and for a fixed energy input of 100 kJ/kg for proteins, and 20 kJ/kg for carbohydrates, respectively. In contrast, while significant differences ( $p \leq 0.05$ ) in the protein content were detected when PEF treatments were carried out at different energy inputs (Fig. 5a), regardless of the field strength applied, only a slighter effect of the energy input was

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1420 431 observed for the extraction of carbohydrates, which was significant ( $p \leq 0.05$ ) only when the energy  
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1422 432 input was increased from 20 to 60 kJ/kg at 10 kV/cm and between 20 and 100 kJ/kg at 30 kV/cm  
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1424 433 (Fig. 5b).

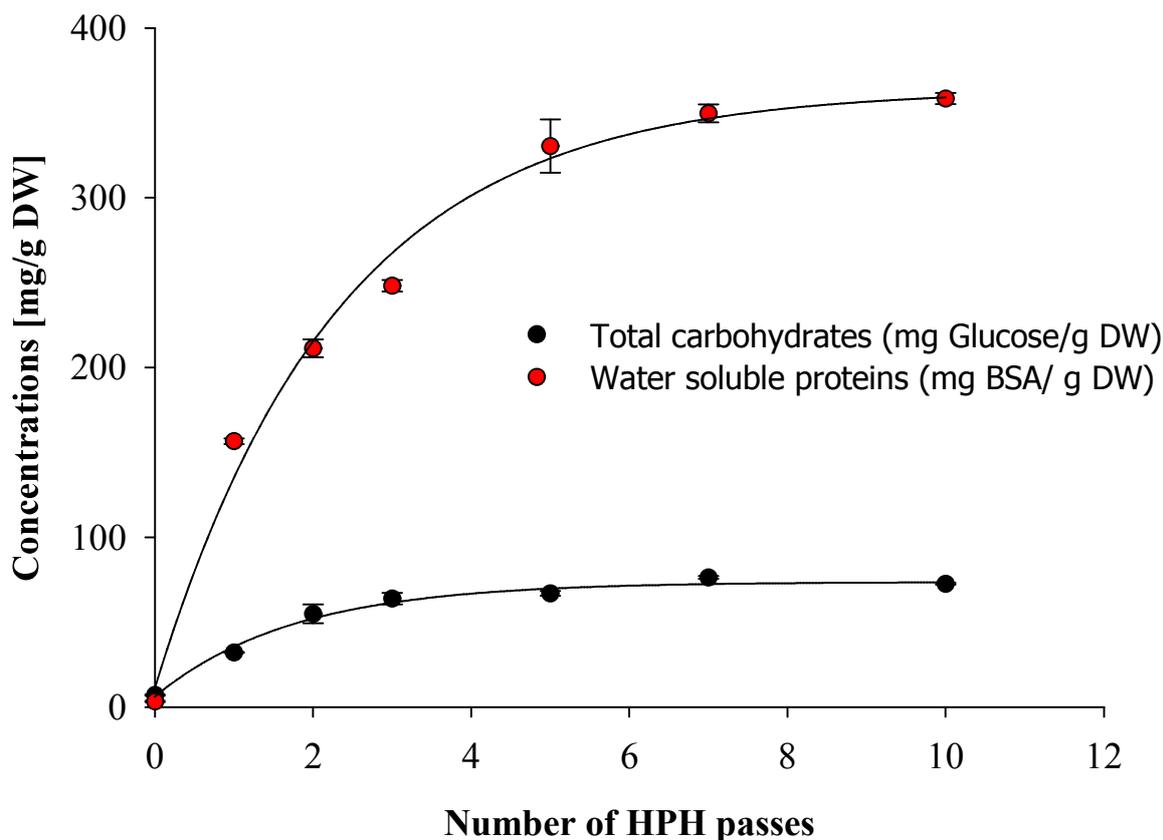
1425  
1426 434 A slightly increasing trend when increasing the energy input from 50 to 150 kJ/kg was previously  
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1428 435 observed by both Goettel et al. [23] with the microalgae *A. protothecoides* at a fixed field strength  
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1430 436 applied of 34 kV/cm, and Pataro et al. [13] with the microalgae *C. vulgaris* at a fixed field strength  
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1432 437 applied of 27 kV/cm. Postma et al. [6], instead, did not find any significant difference in the release  
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1434 438 of carbohydrates from *C. vulgaris* treated by PEF at 50 and 100 kJ/kg at 17.1 kV/cm.

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1436 439 From the results of Fig. 5 it can be concluded that a field strength of 20 kV/cm and an energy input  
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1438 440 of 100 kJ/kg could be sufficient to achieve efficient protein and carbohydrates extraction by PEF.

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1440 441 In particular, assuming a carbohydrates and protein content of 16% and 61 % on DW, respectively  
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1442 442 [6], the amount of these compounds released after PEF treatment (20 kV/cm, 100 kJ/kg) was 35.8%  
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1444 443 (w/w) of total carbohydrates (approximately 5.7% DW biomass) and 5.2% (w/w) of total proteins  
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1446 444 (approximately 3.2% DW biomass). These values are in the same range of values reported by other  
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1448 445 authors [6, 12, 13, 22, 28]. In the study of Postma et al. [6], for example, it was observed that the  
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1450 446 application of a PEF treatment at room temperature resulted in an extraction yield of 22-24% for  
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1452 447 carbohydrates, and 3.2-3.6% for proteins, when the energy input was increased between 50 and 100  
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1454 448 kJ/kg at a field strength applied of 17.1 kV/cm. Moreover, no further improvement of the diffusion  
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1456 449 kinetics of intracellular compounds was detected when PEF effect was combined with the thermal  
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1458 450 treatments at a higher temperature [6] or elevated pH [22].

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1460 451 These results suggest that PEF was successful in opening pores on membranes of *C. vulgaris* cells  
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1462 452 (Figs. 1, 3), allowing the selective release of carbohydrates and small-sized cytoplasmic proteins,  
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1464 453 while hindered simultaneously the diffusion of most proteins, which are likely larger and more  
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1466 454 bonded to the cell structure. This hypothesis is supported by some literature evidence. In fact, it has  
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1468 455 been reported that the proteins of *C. vulgaris* species have molecular weights ranging from 12 to 120  
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1470 456 kDa [26], and that PEF was able to selectively enhance only the extraction of small protein materials,  
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1479 457 with molecular weight lower than 20 kDa, while larger molecules remained entrapped inside the cells,  
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1481 458 being unable to cross the permeabilized cell membrane [6]. In contrast, as suggested by the SEM  
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1483 459 images (Fig. 3), PEF merely electroporated the algae cells without altering the extremely resistant  
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1485 460 rigid cell wall of *C. vulgaris*, which represents a further barrier against the extraction of proteins [46].  
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1487 461 Moreover, it is estimated that 20% of *C. vulgaris* proteins are bonded to the cell wall [47], and  
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1489 462 therefore they likely remained entrapped in the pellet along with the water-insoluble fraction of  
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1491 463 proteins. This would contribute to further explain the relatively low amount of proteins released after  
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1493 464 PEF (Fig. 5b).  
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1495 465 Therefore, the disruption of the rigid cell wall of *Chlorella vulgaris* appears to be a crucial step to  
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1497 466 enhance the protein release [48], hence requiring a more effective cell disruption techniques, such as  
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1499 467 high pressure homogenization [8].  
1500  
1501 468 Fig. 6 reports the amount of carbohydrates and proteins released upon the application of HPH  
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1503 469 treatment (150 MPa) as a function of the number of passes. In agreement with the results of Fig. 1d,  
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1505 470 a significant fraction *C. vulgaris* cells was already disrupted after 1 pass and water gained the access  
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1507 471 to the cytoplasmatic content, allowing the release of a certain amount of carbohydrates and proteins.  
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**Fig.6.** Concentration of proteins and carbohydrates in the surpenatant of untreated ( $n_p=0$ ) and HPH ( $P = 150$  MPa) treated *C. vulgaris* suspension as a function of the number of passages.

The subsequent HPH passes led to the further release of carbohydrates and proteins, whose amount gradually increased up to reaching a saturation value after 5 passes, which was, with respect to the control sample, 9-fold higher for carbohydrates and 200-fold higher for proteins.

An asymptotic behavior in the extraction yield of intracellular compounds, such as chlorophyll and carotenoids, as a result of the increased degree of cell disruption with increasing the number of passes has previously been shown by Xie et al. [49]. These authors reported that the release of these pigments from HPH-processed *Desmodesmus* microalgae could be enhanced by increasing the number of passes up to a saturation value above which no additional leakage of interest compounds could be achieved.

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485 From the results of Fig. 6, using the same assumption for the composition of *C. vulgaris* cells used  
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486 for PEF [6], the amount of carbohydrates and proteins released after 5 HPH passes was 41.9% (w/w)  
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487 of total carbohydrates (approximately 6.7% DW biomass) and 54.1% (w/w) of total proteins  
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488 (approximately 33.0% DW biomass).

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489 Despite the different HPH system and microalgae strain used, these results seem to be consistent with  
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490 those obtained in previous works investigating the extraction of proteins and carbohydrates. In  
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491 particular, Safi et al. [26, 48] found that, among the different cell disruption techniques, including the  
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492 chemical treatments, ultrasonication, and manual grinding, HPH was the most efficient one, and that  
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493 after an HPH treatment ( $P=270$  MPa,  $n_p=2$ ) water gained rapid access to the cytoplasmic proteins and  
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494 infiltrated the chloroplast to recover 50-66% of proteins from the total protein content of *C. vulgaris*  
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495 cells. To date, only Shene et al [27] investigated the effect of HPH process conditions (75-230 MPa,  
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496  $n_p=1-6$ ) on the recovery of total sugars and only from *Nannochloropsis oculata*, whose concentration  
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497 ranged between 55.0 to 62.5mg/g.

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498 However, despite the high cell disruption efficiency of the HPH treatment, the complete release of all  
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499 the proteins contained in the algae could not be reached, because of the rigidity of the cell wall [51],  
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500 as well as the insoluble nature of some proteins that remained in the pellet [50].

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501 The comparison between the results of Figs. 5 and 6 highlight the capacity of PEF to efficiently  
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502 release low molecular weight molecules, such as carbohydrates, to an extent comparable to the one  
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503 obtained from HPH treatment for a sufficiently high number of passes (85.4%). This selectivity of  
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504 PEF towards the carbohydrates could be advantageously exploited for specific applications [41]. In  
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505 contrast, despite the huge increase in protein extraction caused by PEF processing with respect to  
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506 untreated microalgae suspension, the protein yields are still relatively low being 10 fold lower than  
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507 that detected in HPH treated samples.

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508 However, next to the extraction yield of valuable intracellular compounds, the feasibility of a cell  
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509 disintegration technique should also take into account the total energy consumed. In this work, to  
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510 enable the comparison between PEF and HPH, on the basis of the work of Günerken et al. [7], the

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1655 511 total energy consumed (in kWh/kg<sub>DW</sub>) was calculated as the energy to disrupt 1 kg of dry microalgae  
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1657 512 biomass (= consumed energy / (treated biomass · cell disruption yield)), considering a cell disruption  
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1659 513 yield of, respectively, 100% for 5 passages HPH treatment ( $Z_p=1$ ), and 81% ( $Z_p=0.81$ ) for PEF  
1660  
1661  
1662 514 treatment (20 kV/cm, 100 kJ/kg). For HPH, an overall efficiency of the pumping system of 87% was  
1663  
1664 515 considered.

1665  
1666 516 The results showed that HPH is always an extremely energy intensive cell disintegration technique,  
1667  
1668 517 with a total consumed energy 20.0 kWh/kg<sub>DW</sub>, whereas PEF, despite the lower yields is characterized  
1669  
1670 518 by a total consumed the energy of 2.9 kWh/kg<sub>DW</sub>. These results are in contrast with the findings of  
1671  
1672 519 Safi et al.[28], who demonstrated that PEF was energetically less efficient (10.42 kWh/kg<sub>DW</sub>) than  
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1674 520 HPH (0.32 kWh/kg<sub>DW</sub>) after only one passage at 100 MPa when applied for the recovery of proteins  
1675  
1676 521 from *Nannochloropsis gaditana* cells. Probably, this difference can be explained in terms of the  
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1678 522 peculiarity of the tested microalga as well as on the different PEF and HPH systems. For example, in  
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1680 523 agreement with previous findings [52], it is likely that the energy efficiency of the continuous flow  
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1682 524 PEF system used in the present work is higher than that of a batch one used in the work of Safi et al.  
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1684 525 [28].

1685 526 Moreover, for the first time, the comparison between PEF and HPH has also been carried out in terms  
1686  
1687 527 of the energy consumed to extract 1 kg of carbohydrates or proteins, which were, respectively, 40.5  
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1689 528 kWh/kg of glucose equivalent and 72.3 kWh/kg of BSA equivalent for PEF, and 311.8 kWh/kg of  
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1691 529 glucose equivalent and 60.4 kWh/kg of BSA equivalent for HPH. These estimated energy  
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1693 530 consumptions clearly show that the carbohydrates can be efficiently recovered through PEF at  
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1695 531 comparable yields with HPH, but with higher purity and lower energy consumption. This is a  
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1697 532 remarkable result, because a selective release of carbohydrates may result in a less intensive  
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1699 533 fractionation in the later biorefinery stages.

1700 534 In the case of proteins, instead, HPH is slightly more energetically efficient than PEF, because of  
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1702 535 significantly higher yields. However, PEF represents a viable option when considering the lower  
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1704 536 purity of HPH extracts and the need of more complex downstream purification process. In addition,  
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537 PEF and HPH significantly differ also in the composition of the protein extracts, and therefore future  
538 research should address in deeper details the effect of microalgae pretreatment on the molecular  
539 composition of the protein extract.

#### 4. Conclusions

The present study provides additional insights into the impact of PEF and HPH treatments on the disintegration efficiency of *C. vulgaris* cells and into the subsequent recovery of intracellular compounds, namely carbohydrates and proteins.

PEF resulted in being a relatively mild cell disruption method, which merely electroporates the algae cells without the formation of any cell debris, allowing to selectively enhance the extraction yield of small ionic substances and carbohydrates to an extent comparable to that achieved by HPH. The extraction efficiency of proteins, instead, was relatively low and did not exceed 5.2% of the total.

HPH, instead, was able to disrupt completely the microalgae cells, favoring an instantaneous and efficient release of all the intracellular material, including a large amount of proteins, whose release was 10.3 fold higher than by PEF. However, despite the higher extraction efficiency, the formation of large amounts of finely sized cell debris by HPH significantly complicates any downstream separation process.

Moreover, the HPH treatment resulted in being significantly more energy-intensive than PEF to achieve a comparable release of carbohydrates, while shown a slightly higher energy efficiency when used for the extraction of proteins.

In the ongoing work, the optimal cell disruption conditions identified in this work for individual PEF ( $E = 20 \text{ kV/cm}$ ;  $W_T = 100 \text{ kJ/kg}_{\text{SUSP}}$ ) and HPH ( $n_P = 5$ ) treatment, are tested in a cascade biorefinery, in order to maximize in a selective and sustainable way the extraction yield of target compounds, by reducing the overall processing costs, which nowadays represent the main bottleneck to the full exploitation of microalgal biomass.

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## **Acknowledgements**

Authors wish to thank Eng. Luigi Esposito and Dr. Maria Rosa Scognamiglio for their invaluable help with particle size distribution and SEM analyses, respectively. This research did not receive any specific grant from funding agencies in the public, commercial, or not-for-profit sectors.