1	Simultaneous disinfection and microcontaminants elimination of urban wastewater					ter		
2	secondary effluent by solar advanced oxidation sequential treatment at pilot scale							
3	Gulnara	Maniakova ^a ,	María	Inmaculada	Polo-López ^{b,c} ,	Isabel	Oller ^{b,c} ,	Sixto
4	Malato ^{b,c}	**Luigi Rizzo ^{a*}						

- 5
- ^aWater Science and Technology (WaSTe) group, Department of Civil Engineering, University of
 Salerno, Via Giovanni Paolo II 132, 84084 Fisciano, SA, Italy.
- 8 ^bCIEMAT-Plataforma Solar de Almería. Ctra. Senés km 4, 04200 Tabernas (Almería), Spain.
- 9 ^cSolar EnergyResearch Centre (CIESOL), Joint Centre University of Almería-CIEMAT,
 10 Carretera de Sacramento s/n, E-04120, Almería, Spain.
- 11
- 12 Pre-review version of the manuscript published in Journal of Hazardous Materials (436 (2022)
- 13 129134, https://doi.org/10.1016/j.jhazmat.2022.129134).
- 14
- 15 *Corresponding Author. Phone:+39089969334
- 16 ** Corresponding Author. Phone:+34950387940
- 17 E-mail addresses: <u>l.rizzo@unisa.it</u>(Luigi Rizzo), <u>smalato@psa.es</u> (Sixto Malato)
- 18

19 ABSTRACT

Simultaneous contaminants of emerging concern (CECs) removal and bacteria inactivation by 20 sequential treatment sunlight/H₂O₂ and solar photo-Fenton (SPF) with EDDS at near neutral pH 21 22 was investigated. Process efficiency was evaluated in terms of (i) degradation of five CECs (namely caffeine, carbamazepine, diclofenac, sulfamethoxazole and trimethoprim) at the initial 23 concentration of 100 µgL⁻¹ each and (ii) bacteria inactivation (Escherichia coli (E. coli) 24 and Salmonella spp). The effect of H₂O₂, Fe and EDDS concentration and Fe:EDDS dosing 25 time was evaluated. 60 % removal of the sum of total CECs and pathogens inactivation below 26 the detection limit (DL) were observed by the sequential treatment with Fe:EDDS additions at 27 60 min and 45 min in simulated urban wastewater effluent. CECs degradation and bacteria 28 inactivation rate were higher with Fe:EDDS 0.1:0.1 mM than with Fe:EDDS 0.05:0.05 29 mM. Sequential treatment was validated in actual urban wastewater effluent, being able to 30 remove 60% of the target CECs and inactivate bacteria below the DL. Increasing EDDS 31 concentration negatively affected Salmonella spp inactivation. Sequential treatment based on 32 120 minutes of sunlight/H₂O₂ (50 mg L^{-1}) and subsequent SPF with Fe:EDDS (0.1:0.1 mM) was 33 chosen as best operation conditions for full scale treatment in urban wastewater treatment plants. 34

35

36 Keywords: Advanced Oxidation Processes, contaminants of emerging concern, *Escherichia*37 *coli*, *Salmonella spp*, , wastewater reuse

39 1 Introduction

Modern society is concerned about water scarcity and pollution of water resources. World 40 population growth, economic development and climate change entail a growing demand for 41 42 water resources that can only be met through the use of unconventional water sources. Treated urban wastewater are considered a necessary alternative to freshwater resources (Scheierling et 43 al., 2011) and several studies have outlined the importance of urban wastewater treatment plant 44 effluent (UWWE) reuse in agriculture, industry, urban development, domestic, potable water 45 supply and other applications (Lahlou et al., 2021; Leonel and Tonetti, 2021; Liao et al., 46 2021). Urban wastewater treatment by conventional methods cannot effectively remove some 47 challenging pollutants including contaminants of emerging concern (CECs) (pesticides, 48 pharmaceuticals, personal care products, etc.) and pathogens (bacteria, viruses, protozoa and 49 parasites) that can be easily taken up by plants irrigated with UWWE (Wu et al., 2014) and lead 50 to toxic effects in humans (Delli Compagni et al., 2020; Fent et al., 2006). 51

One suitable option to improve the efficiency of conventional urban wastewater treatment plants 52 (UWWTPs) is their upgrading with advanced/tertiary treatment methods such as Advanced 53 Oxidation Processes (AOPs), which are based on the formation of highly reactive and non-54 selective hydroxyl radicals (HO') that can effectively remove CECs as well as inactivate 55 microorganisms. AOPs operated with solar radiation make the process more sustainable by 56 saving energy cost (Malato et al., 2009). Solar photo-Fenton (SPF) has been successfully 57 investigated for CECs removal and bacteria inactivation (Giannakis et al., 2017). Nevertheless, 58 in spite of the high efficiency of such process, a main disadvantage is the necessity of adjusting 59 pH to 2.8 to avoid iron precipitation and subsequent effluent neutralization before discharge or 60 61 reuse (Pignatello et al., 2006). Using mono-, poly-, or amino-carboxylic acids such as Ethylenediamine-N, N'-disuccinic acid (EDDS) allows to solve these problems and speed up 62 their full scale application. Fe:EDDS complex formation avoids iron precipitation and makes the 63 process effective even at neutral pH (Zhang and Zhou, 2019). However, the efficiency of SPF 64

with EDDS for the inactivation of bacteria seems to be contradictory, with regard to the higher bacteria resistance in the presence of organic matter due to the EDDS dosing(García-Fernández et al., 2019). The results observed in scientific literature conclude that, though sunlight/H₂O₂ can easily and efficiently inactivate different microorganisms usually found in UWWE mainly due to internal photo-Fenton mechanism (Giannakis, 2018), it cannot produce sufficient radicals to effectively degrade CECs . Simultaneous removal of CECs and bacteria inactivation has received only poor attention so far (Soriano-Molina et al., 2019a).

In a recent work, Maniakova et al., (2021b) compared the efficiency of SPF with EDDS and 72 sunlight/H₂O₂ processes separately for the simultaneous CECs removal and bacteria inactivation. 73 They confirmed that sunlight/H₂O₂ is more effective than SPF with EDDS for bacteria 74 inactivation and is not able to remove CECs. On the other side, they observed high efficiency of 75 SPF with EDDS for CECs removal. Therefore, aiming to achieve a complete bacteria inactivation 76 and, at least, 60% removal of CECs only in the tertiary step, sequential treatment by 77 sunlight/H₂O₂ process and SPF with EDDS could be considered a suitable option. The aim to 78 achieve 60 % removal of the sum of total CECs only in the tertiary treatment is based on the 79 Switzerland law in which 80% removal of microcontaminants is requested along the whole train 80 in the UWWTP (from primary to tertiary treatment). 81

Effluents from municipal wastewater treatment plants disinfection and CECs removal by solar driven AOPs (namely sunlight/H₂O₂ and SPF, separately) have been successfully investigated in low cost Raceway Pond Reactors (RPRs). RPR allows to reduce the treatment cost in this type of mild effluent in comparison with a tubular photo-reactor equipped with compound parabolic collectors (CPCs), commonly used in photocatalytic processes (Cabrera-Reina et al., 2021).

87 Treatment time plays a major role in full-scale UWWTPs. The minimization of treatment time 88 for a specific treatment goal allows keeping the reactor solar collecting surface as small as 89 possible, decreasing treatment costs. Therefore, the main goal of this work was to investigate the 90 capability of a sequential treatment using sunlight/ H_2O_2 followed by SPF, for guarantying both

disinfection and elimination of CECs, with EDDS in solar RPRs for tertiary treatment of UWWE 91 at near neutral pH to minimize treatment time. The removal of a mixture of CECs as caffeine 92 (CAF), carbamazepine (CBZ), diclofenac (DCF), sulfamethoxazole (SMX) and trimethoprim 93 94 (TMP) and bacteria (E. coli and Salmonella) inactivation were monitored in both simulated and actual UWWE. To authors' knowledge, the effect of different times of Fe:EDDS dosing and 95 H₂O₂ as well as different Fe:EDDS concentrations in the sequential treatment sunlight/H₂O₂ 96 followed by sunlight/H₂O₂/Fe:EDDS at circumneutral pH was investigated for the first time 97 focusing both CECs and bacteria. 98

- 99
- 100 2 Materials and methods

101 *2.1 Chemicals and reagents*

Selected model CECs were all of high-purity grade (>99%): CAF was provided by Fluka and CBZ, DCF, SMX and TMP were purchased from Sigma Aldrich. Fe₂(SO₄)₃·H₂O 75% solution (Panreac) was used as Fe(III) source. EDDS water solution (35% w/v), H₂O₂ (35%, w/v), bovine liver catalase, acetonitrile (ACN) (UHPLC-grade) and formic acid (UHPLC-grade) were provided by Sigma Aldrich. Titanium(IV) oxysulfate (Sigma-Aldrich) was used for H₂O₂ measurements. Reagents for dissolved iron determination (namely,1,10-phenanthroline, ammonium acetate, ascorbic acid) were obtained from Sigma-Aldrich.

109 *2.2 Water matrices*

110 Two water matrices, simulated urban wastewater effluent (SUWWE) and UWWE, were used in 111 the experiments. The main physicochemical properties measured for both water matrices are 112 shown in Table 1. SUWWE was prepared using the following chemicals and 113 concentrations (Sánchez-Montes et al., 2020): *i*) organic matter: sodium lignin sulfonate (2.4 mg 114 L^{-1}), humic acid (4.2 mg L^{-1}), sodium lauryl sulphate (0.9 mg L^{-1}) (Sigma-Aldrich); peptone (2.7 115 mg L^{-1}) and beef extract (1.8 mg L^{-1}) (Biolife); acacia gum powder (4.7 mg L^{-1}) and tannic acid 116 (4.2 mg L^{-1}) (Panreac); *ii*) inorganic salts: 23.6 mg L^{-1} (NH₄)₂SO₄ and 60 mg L^{-1} CaSO₄·2H₂O

- 117 (from Panreac); 4 mg L⁻¹ KCl (from J.T. Baker); 7.0 mg L⁻¹ K₂HPO₄, 96 mg L⁻¹ NaHCO₃, 580
- 118 mg L^{-1} NaCl and 60 mg L^{-1} MgSO₄ (Sigma-Aldrich).
- 119 Several samples of UWWE were freshly collected after secondary treatment and sand filtration of 120 the urban UWWTP of Almería, El Bobar (Spain). The water samples were stored at 4 °C not 121 more than 3 days and characterized before each use (Table 1). The samples showed a significant 122 high carbonates concentration (446.0±82.5 mg L⁻¹: mainly as HCO_3^{-}), a known scavengers 123 of HO[•], therefore the water was previously adjusted to <75±6 mg L⁻¹ through the addition of 124 sulfuric acid. During the carbonates stripping, pH did not change significantly (it was in the range 125 of neutrality, 6.9–7.9) and did not affect the naturally occurring microbial population.
- 126 Table 1 SUWWE and UWWE characterization.

Parameter	SUWWE	UWWE	
pH	7.6±0.3	7.9±0.1	
Conductivity (mS/cm)	$1.4{\pm}0.1$	2.6 ± 0.1	
Turbidity (NTU)	3.4±0.2	8.3±2.6	
*DOC (mg L ⁻¹)	15.5±0.6	20.0 ± 2.6	
HCO_3^{-} (mg L ⁻¹)	67.5.±6.0	446.0±83**	
$Cl^{-}(mg L^{-1})$	355±9	547±10	
NO_{3}^{-} (mg L ⁻¹)	5.4 ± 0.2	17.5 ± 1.0	
PO_4^{3-} (mg L ⁻¹)	6.1±0.3	38.9±1	
SO_4^{2-} (mg L ⁻¹)	119±5	138.0±2	
$NH_{4^{+}}(mg L^{-1})$	6.4 ± 0.1	35.6±1.2	
$Na^+(mg L^{-1})$	256±10	275±9	
$K^{+}(mgL^{-1})$	14.8 ± 1.1	28.0 ± 2.5	
$Ca^{+}(mg L^{-1})$	27.8 ± 1.7	100±2	

127 *DOC – dissolved organic carbon; **adjusted to around $75\pm6mg L^{-1}$ before the treatment.

128

- 129 CECs mixed stock solution was prepared in methanol at 2.5 g L^{-1} of each CEC. Organic 130 contribution of methanol was DOC = 12 mg L^{-1} . Further increase of DOC (12 mg L^{-1}) was also 131 observed after the addition of 0.1 mM of EDDS.
- 132 *2.3 Analytical measurements*

133 The concentrations of Fe and H_2O_2 were measured spectrophotometrically by an UV-Vis 134 Evolution 220 spectrophotometer (Thermo scientific). H_2O_2 concentration was monitored 135 according to DIN 38402H15 method at 410 nm. Fe concentration was analyzed by 1,10-136 phentranoline following ISO 6332 method at 510 nm. All samples were filtrated by 0.45 μ m 137 nylon filter before the analysis. Temperature (Thermometer, HANNA), pH (GLP 22 pH meter, 138 CRISON), turbidity (2100N Turbidimeter, HACH) and conductivity (GLP 31 Conductimeter,
139 CRISON) were also monitored. DOC and carbonates were measured using a TOC-VCSN
140 analyzer (Shimadzu) in filtered samples (0.45 µm nylon filter).

CECs concentrations were analyzed by ultra-performance liquid chromatography (UPLC Agilent 141 Technologies, Series 1200) through UV-DAD detector, Poroshell 120 EC-C18 column (Agilent 142 Technologies: 50 mm \times 3.0 mm, 2.7 µm particle). The simultaneous analysis of the 5 CECs was 143 done according to previous reported working conditions (Maniakova et al., 2022). Briefly, the 144 initial 100% aqueous (formic acid 25 mmolL⁻¹) solution was varied in 10 min up to achieve 50% 145 of ACN, and 100% of ACN in the subsequent 2 min. 9 mL of collected sample was filtered using 146 a 0.22 µm PTFE filter (Millipore), and mixed with 1 mL of ACN used for washing the filter to 147 remove any adsorbed CEC. 148

149 2.4 Bacterial quantification analysis

E. coli O157:H7 (CECT 4972) and S. enteritidis (CECT 4155) (provided by the Spanish Culture 150 151 Collection (CECT) as freeze-dried cultures) were used for SUWWE tests. E. coli and S. enteritidis strains were inoculated in 14 mL of Nutrient Broth (a mixture of NaCl, Beef extract, and 152 Peptone), and Tryptone Soya Broth (OXOID), respectively, and grown aerobically in a rotary 153 shaker (90 rpm) at 37°C for 20 h. Cells were harvested by centrifugation at 3000 rpm for 15 min 154 (J.P. Selecta) and the pellet was re-suspended in 14 mL phosphate-buffer saline solution 155 (PBS, Oxoid), yielding a final concentration of 10⁹ CFU mL⁻¹.An aliquot of each bacterial 156 suspension was added in SUWWE to obtain an initial concentration of 10³ CFU mL⁻¹. 157

Indigenous *E.coli* and *Salmonella spp* were analysed in tests with freshly collected UWWE. Enumeration of bacteria was performed by standard plate counting method using selective agar media: Chromocult[®] (Merck) for *E.coli* and Salmonella Shigella agar (Scharlau) for *Salmonella spp*. Water samples (50-500 μ L) were spread onto each corresponding selective agar Petri dish. Subsequently, plates were incubated for 24 h (*E.coli*) and 48 h (*Salmonella spp*) at 37 °C and counted. The detection limit (DL) in SUWWE was 200 CFU 100 mL⁻¹. In UWWE, when bacterial concentration was expected to be lower than 200 CFU 100 mL⁻¹, samples were processed by the membrane filtration method. For each bacteria species, 100 mL of sample were filtered using a 0.45 μ m-pore-size cellulose nitrate membrane (Sartorius) and a Microfil Filtration System (Millipore) (Sánchez-Montes et al., 2020). Then, the obtained membranes were plated in the corresponding medium. DL of this technique is 1 CFU 100 mL⁻¹, to take into account the limit for class A treated wastewater (10 CFU of *E. coli*100 mL⁻¹) set by the new European Regulation on minimum requirements for water reuse ((EU) 2020/741).

171 To avoid any potential effect of residual H_2O_2 over bacterial viability, the residual H_2O_2 was 172 quenched in the sample using bovine liver catalase (Sigma-Aldrich) (1 mL sample was mixed 173 with 20 μ L of catalase stock solutions at 0.1 g L⁻¹) (Sánchez-Montes et al., 2020).

174 2.5 Experimental set-up

175 Treatment tests were carried out in a RPR at pilot plant scale under natural solar 176 irradiation at Plataforma Solar of Almeria, Spain, in clear sunny days. The five target CECs were 177 added at an initial concentration of 100 μ gL⁻¹ each one from a mix stock solution. *E. coli* and *S.* 178 *enteritidis* were spiked with SUWWE at an initial concentration of 10³ CFU mL⁻¹ each one. In 179 UWWE, CECs were spiked and naturally occurring *E. coli* and *Salmonella spp* were analyzed.

The RPR photoreactor used has been described in detail elsewhere (Costa et al., 2020) and briefly 180 it consists of an open reactor with dimensions of 97 x 45 cm, 15 cm liquid depth and a working 181 total volume of 90 L. Prior to start with solar exposure, the aqueous solution was homogenized 182 in the dark, according to the following additions sequence: CECs and bacteria mixture (if 183 necessary) for 10 min and then H₂O₂ for 5 min. After, the reactor was uncovered and experiment 184 started. The desired Fe:EDDS concentration (prepared according to the protocol described in 185 186 Maniakova et al., (2021b)) was added after different times of sunlight/H₂O₂ process. The solar UV-A radiation (data in W/m^2) was measured by a pyranometer (280-400 nm, Model CUV-5, 187 Kipp&Zonen) mounted on a horizontal platform and located close to the RPR at PSA 188 facilities. To compare results from different experiments, the accumulated UV energy per unit 189

190 of treated volume (Q_{UV} , kJ L ⁻¹) was calculated according to the following equation (Eq.2) 191 (Malato et al., 2003):

$$Q_{UV,n} = Q_{UV,n-1} + \Delta t_n \cdot \overline{UV}_{G,n} \cdot \frac{A_r}{V_t}$$
(Eq. 2)

193 where, Q_{UV} is the accumulated UV energy per treated volume between samples *n* and *n*-1. $\overline{UV}_{G,n}$ 194 (W m⁻²) is the average UV radiation measured and Δt_n is the experimental time between samples. 195 A_r is the illuminated area (m²) and V_t is the total volume of water (L).

The sequential treatment was always performed with sunlight/H₂O₂ process as initial step followed by the addition of Fe:EDDS to promote SPF. The performance of the sequential treatment was evaluated by varying the reagents concentration (H₂O₂ and Fe:EDDS) as well as the addition-time of Fe:EDDS (named dosing strategy, Table 2). In all performed experiments H₂O₂ concentration was measured during the treatment and kept close to the initial concentration (when the concentration of H₂O₂ decreased to around 10 mg L⁻¹, H₂O₂ was added to the solution).

203 Table 2 – Sequential treatment operating conditions and Fe:EDDS dosing strategy.

	SUWWE			UWWE			
$[H_2O_2]_0$	[Fe:EDDS] ₀ [Fe:EDDS] ₀		[Fe:EDDS] ₀	[Fe:EDDS] ₀	[Fe:EDDS] ₀		
	0.1:0.1 mM	0.05:0.05 mM	0.1:0.1 mM	0.1:0.2 mM	0.1:0.1 mM - twice		
		Fe:EDDS dosing strategy					
30 mgL ⁻¹	60 min	-	-	-	-		
50 mgL ⁻¹	30, 45 or 60 min	60 min	120 min	150 min	120 & 150 min		
100 mgL ⁻¹	45 min	-	-	-	-		

204

192

205 **3 Results and discussion**

- 206 *3.1 CECs degradation and bacteria inactivation in SUWWE*
- 207 *3.1.1 Effect of Fe:EDDS dosing strategy*
- 208 To assess the best treatment time for simultaneous CECs removal and bacteria inactivation, the
- effect of different Fe:EDDS (0.1:0.1 mM) dosing times (60, 45 and 30 min after H₂O₂) were
- initially investigated with sunlight/ H_2O_2 at 50 mg L⁻¹ in SUWWE (Fig. 1).



Sunlight/H₂O₂ disinfection efficiency was reduced as dosing time was decreased from 60 to 30

min (Fig. 1a, c, e). The effect of sunlight/H₂O₂ process on CECs degradation was poor for all

211

Figure 1–Bacteria inactivation and CECs degradation in SUWWE by sequential treatment with sunlight/H₂O₂ (H₂O₂=50 mg L⁻¹) with subsequent addition of Fe:EDDS (0.1 mM Fe:EDDS) at several dosing times: 60 min (a, b), 45 min (c, d) and 30 min (e, f) in RPR.

220

The inactivation of bacteria observed is attributed to the accumulated internal cell damages due 221 to sunlight/H₂O₂ process. The internal cell damages are attributed to: i) the formation of 222 223 reactive oxygen species (ROS) (direct action) by UV solar photons; ii) intracellular photo-224 Fenton-like reactions between H₂O₂, which penetrates through the bacterial cell wall, and naturally occurring intracellular-iron (indirect action), generating HO[•]. Other ROS directly 225 226 attack susceptible moieties (oxidative stress) or are involved into the reduction of iron. Added H₂O₂ also initiate bacterial membrane auto-oxidation (outer damage). Light reduced ferric iron 227 to ferrous initiating a photo-catalytic cycle. These processes affect different intracellular vital 228 components leading to bacterial death or lack of viability (Giannakis et al., 2016). 229

Nevertheless, in our experimental conditions, the variation of the treatment time showed 230 231 different inactivation behavior, and the susceptibility of both bacteria to this treatment was also different. E. coli showed a faster abatement than S. enteritidis in all cases, which is in agreement 232 with previous findings (Maniakova et al., 2021b; Nahim-Granados et al., 2018). Higher 233 resistance of S. enteritidis can be explained by its capability to adapt to sunlight stress. Some 234 investigations suggest that the gene *RpoS*, which controls the expression of genes involved in 235 prevention of oxidative damage, could be relevant in the photodynamic action of the radiation 236 in S. typhimurium (Gómez-López et al., 2014). In spite of the differences between both bacteria, 237 the solar irradiance time in combination with H₂O₂ plays a major role, indicating that 60 min 238 239 of solar exposure is required to guarantee a complete inactivation of E. coli (DL= 200 CFU 100 mL⁻¹); while the DL was not reached for S. *enteritidis* during the sunlight/H₂O₂ process. 240 The kinetics profiles obtained during sunlight/H₂O₂ process in all cases showed a first stage 241 where the cell damages are being accumulating (shoulder phase) with a log-decay occurring 242

after that. The shoulder phase required 30 min of solar treatment for E. coli, while it was less 243 evident for S. enteritidis. The oxidative damages accumulation over time was necessary to 244 overcome the capability of self-defense of the cells to finally induce the cell death. During the 245 246 subsequent SPF phase, the residual bacterial cells further decreased till the DL, except for S. enteritidis with Fe:EDDS and dosing time at 30 min. The bacterial inactivation behavior 247 showed that the higher exposure time during the first phase (sunlight/H₂O₂), the faster 248 249 inactivation in the second phase (SPF) in terms of treatment time. This effect could be explained 250 by the sub-lethal damage state on the remaining viable cells occurring during the first phase, making them more vulnerable for the potential external damages generated by early minutes of 251 252 SPF. Nevertheless, the particular case of S. enteritidis where the DL was not reached with Fe:EDDS dosing time at 30 min, indicated that the production of HO' by SPF was not sufficient 253 and that others boundary and protecting effects over bacteria occurred. In fact, this limited effect 254 of SPF with EDDS for bacterial inactivation has been previously described, and can be possibly 255 attributed to: i) the iron precipitation formed particles that may act as screen protecting 256 257 bacteria from the action of solar UVA photons, and *ii*) the organic matter concentration from a biodegradable substance (EDDS) that made possible a more favorable ambient for 258 bacteria survival 259

260 The H₂O₂ concentration measured during sunlight/H₂O₂ did not show any significant change, which is expected due to the no O-O bond breaking (required for HO' generation) under natural 261 solar radiation (Giannakis et al., 2016). The absence of HO' generation is the main reason of 262 the low CECs degradation rate observed during this phase, except in the case of DCF which 263 suffered photolysis, reaching 59, 36 and 30% removals, before Fe:EDDS was added after 60, 264 45 and 30 min of sunlight/H₂O₂, respectively. These results are in agreement with a previous 265 work in which a poor removal of CBZ and SMX (only 20 % and 17 %, respectively) was 266 observed by sunlight/H₂O₂, while DCF was efficiently photolysed (Moreira et al., 2018). Under 267 SPF, similar CECs degradation kinetics were observed in all experiments (Fig. 1 b, d, 268

f) independently of the Fe:EDDS adding time. The aim of 60 % removal of the sum of total 269 CECs was achieved after 5 min of Fe:EDDS addition. The highest elimination rate was 270 observed in the first 15 minutes after Fe:EDDS addition, then the reaction rates slowed down 271 272 and the residual concentration of the CECs did not significantly change till the end of the experiment. When Fe:EDDS was added, regardless of dosing time, a rapid consumption of 273 almost 50% of H₂O₂ was observed. Such H₂O₂ consumption after Fe:EDDS addition can be 274 explained by the degradation of the complex and Fe^{2+} release after the solar irradiation of 275 Fe³⁺:EDDS. Fe²⁺ reacts with H₂O₂ engendering to HO[•] and Fe³⁺, which precipitates as Fe(OH)₃ 276 at circumneutral pH. Iron precipitation also explains the incomplete degradation of CECs. 277

278 The initial dissolved Fe concentration decreased during the treatment. Final Fe concentration was 0.04, 0.06 and 0.05 mM in the end of the sequential treatment with Fe:EDDS at 60, 45 279 and 30 min dosing times, respectively. Such high Fe precipitation means that Fe:EDDS was no 280 longer stable and/or the Fe:EDDS was degraded by radicals in the treatment of SUWWE at 281 circumneutral pH (Soriano-Molina et al., 2019). The Fe:EDDS complex yields the Fe:EDDS 282 radical (reaction 1), promoting the generation of HO[•] and O₂[•] under natural solar radiation 283 (reactions 2, 3). As Fe:EDDS degradation starts, a major part of Fe precipitates, but a small 284 amount could promote other photo-Fenton cycle reactions, or Fe:EDDS could be regenerated 285 286 when more EDDS is available in the solution (Miralles-Cuevas et al., 2019).

287
$$\operatorname{Fe}^{3+}\operatorname{EDDS} + \operatorname{hv} (290-500 \text{ nm}) \rightarrow [\operatorname{Fe}^{3+}\operatorname{EDDS}]^* \rightarrow \operatorname{Fe}^{2+}\operatorname{EDDS}^{*3-}$$
(1)

$$EDDS' + O_2 \rightarrow O_2' + EDDS^{2-}$$
(2)

$$EDDS'+OH^{-} \rightarrow EDDS^{2-} + HO'$$
(3)

290

Summarizing, the best operating conditions for the removal goal of CECs and bacteria (DL = 200 CFU 100 mL⁻¹), were obtained after 75 min of sequential treatment time considering 60 min sunlight/H₂O₂ followed by 15 min of SPF.

With the aim to evaluate the effect of initial H₂O₂ concentration on the efficiency of the 296 sequential treatment, different H₂O₂ doses were investigated. The results of the sequential 297 treatment with sunlight/H₂O₂ at 100 mgL⁻¹ and Fe:EDDS (0.1:0.1 mM) dosing after 45 min are 298 presented in Fig. 2. H₂O₂ concentration improved slightly the efficiency of *E. coli* inactivation 299 showing 15 min faster inactivation with H_2O_2 100 mg L⁻¹ in comparison with 50 mg L⁻¹. These 300 results are in agreement with scientific literature. Rodríguez-Chueca et al., (2014) investigated 301 302 *E.coli* inactivation by sunlight/H₂O₂ process with different H₂O₂ concentrations (5-50 mg L^{-1}) in a CPC reactor. These results can be explained through the mechanism of bacteria inactivation 303 304 discussed above. The frequency of oxidative attacks and the concentration of oxidative species are responsible for bacterial inhibition and final destruction, limited by the iron 305 available inside the cells. Therefore, increasing H₂O₂ dose has a threshold (Rodríguez-Chueca 306 et al., 2012). However, the increase of H_2O_2 concentration till 100 mg L⁻¹ improved S. 307 *enteritidis* inactivation, achieving DL 30 min faster, compared to 50 mg L^{-1} H₂O₂. 308

309 On the other hand, lower H₂O₂ initial concentration (30 mg L⁻¹) was less effective for both 310 bacteria inactivation, showing longer time to get the DL: 90 min (Q_{UV} =1.6 kJ L⁻¹) and 105 min 311 (Q_{UV} =1.7 kJ L⁻¹) for *E.coli* and *S. enteritidis*, respectively (Fig. 3).





Figure 2–Bacteria inactivation (a) and CECs degradation (b) in SUWWE by sequential treatment with sunlight/H₂O₂ (H₂O₂=100 mg L⁻¹) and SPF (45 min Fe:EDDS dosing; Fe= 0.1 mM, EDDS=0.1 mM) in RPR.





Figure 3–Bacteria inactivation (a) and CECs degradation (b) in SUWWE by sequential treatment with sunlight/H₂O₂ (H₂O₂ =30 mg L⁻¹) and SPF (60 min Fe:EDDS dosing; Fe= 0.1 mM, EDDS=0.1 mM) in RPR.

324

H₂O₂ concentration did not affect the efficiency of CECs removal showing a similar behavior. 325 The aim to remove 60 % of sum of total CECs was reached 5 minutes after the addition of 326 Fe:EDDS even in the experiment with low H_2O_2 concentration (30 mg L⁻¹). These results are 327 in agreement with a previous work (Dong et al., 2019). They investigated different Fe:EDDS 328 and H₂O₂ concentrations and observed that the removal efficiencies of the target CECs by 329 UVA/Fe/EDDS/H₂O₂ process strongly depended on the Fe:EDDS dose and 330 less on the H_2O_2 dose. H_2O_2 consumptions were 25 mgL⁻¹ and 20 mgL⁻¹ at the end of the treatment 331 for 100 mg L^{-1} and 30 mg L^{-1} of H₂O₂, respectively. The final Fe concentration with 100 mg L^{-1} 332 1 and 30 mgL⁻¹ of H₂O₂ was 0.03 mM and 0.08 mM, respectively. 333

334

335

3.1.3 Effect of decreasing Fe and EDDS concentration

In order to evaluate the effect of Fe and EDDS concentration on the CECs removal, sequential treatment with sunlight/H₂O₂ (50 mg L⁻¹) and SPF with Fe:EDDS 0.05:0.05 mM (60 min dosing time) was investigated (Fig. 4) and compared with the results presented in Fig. 1a,b (Fe:EDDS 0.1:0.1mM). No attempt was done to increase EDDS or Fe concentration up 0.1 mM as Fe > 0.1 mM would not be consistent with reuse of treated water for irrigation or disposal and > 0.1 mM EDDS would increase organic content of treated effluent. The changing in Fe:EDDS concentration mainly affected CECs degradation. *E.coli* and *S. enteritidis* showed similar behavior (data not shown) compared to the results presented on the Fig 1 but 15 min and 30 min more were needed to reach the DL for *E.coli* and *S. enteritidis*, respectively. The differences could be attributed to the lower concentration of Fe:EDDS in reactions 1, 2 and 3.



346

Figure 4– CECs degradation in SUWWE by sequential treatment with sunlight/H₂O₂ (H₂O₂ =50 mg L⁻¹) and SPF (60 min Fe:EDDS dosing; Fe= 0.05 mM, EDDS=0.05mM) in RPR.

The aim of getting 60% removal of total CECs was achieved within 5 min by SPF with Fe:EDDS 0.1:0.1 mM (Fig.1b), but longer solar irradiation time (20 min) was needed in sequential treatment with Fe:EDDS 0.05:0.05 mM. To the best of our knowledge simultaneous water decontamination by SPF with such low Fe and EDDS concentration (0.05 mM) has not been previously investigated. Results showed that low Fe:EDDS provoked not enough elimination of CECs to reach the goal. Therefore, no attempt was done to decrease further the concentration of the complex.

358 *3.2 CECs degradation and bacteria inactivation in UWWE*

359 The capability of the sequential treatment was validated in UWWE in order to evaluate the 360 water matrix effect limitations on the entire processes performance.

Preliminary performed tests following best strategy obtained in SUWWE demonstrated that 361 362 running sunlight/H₂O₂ for 60 min was not efficient for naturally occurring bacteria inactivation in UWWE (Maniakova et al., 2021a). It is well-known that bacteria inactivation in actual 363 UWWE usually requests more time compared to model one due to its inherent physical-364 chemical and microbial complexity containing a variable concentration of different organic and 365 inorganic compounds which can act not only as HO' scavenger but also protecting bacteria 366 from sunlight photons, resulting in an overall lower inactivation efficiency (Rincón and 367 Pulgarin, 2004). Therefore, in the experiment in UWWE the time of the first phase 368 (sunlight/H₂O₂) of the sequential treatment was increased twice (Fe:EDDS dose 0.1 mM was 369 added after 120 min). The results for simultaneous CECs removal and bacteria inactivation by 370 371 sequential treatment (sunlight/H₂O₂ and SPF with Fe:EDDS addition after 120 min) are shown in Fig. 5. The treatment was effective for water disinfection as DL (1 CFU 100 mL⁻¹) was 372 reached for both investigated bacteria after 150 min ($Q_{UV}= 2.1$ kJ L⁻¹) and 210 min ($Q_{UV}= 2.9$ 373 374 kJ L⁻¹) for Salmonella spp and E. coli, respectively.

In SUWWE, S. enteritidis showed higher resistance compared to E. coli. Opposite, in 375 UWWE, Salmonella spp was inactivated faster than E.coli. These results are in agreement with 376 scientific literature. Aguas et al., (2019) investigated inactivation of E. coli and Salmonella spp. 377 378 naturally occurred in UWWE by sunlight/H₂O₂ and by SPF at neutral pH in CPC reactor. They 379 also observed Salmonella spp was inactivated faster than E.coli. Therefore, the different resistance of both bacteria in different water matrices reinforces the currently approach 380 suggested in literature about the need to test different water matrices to determine the true 381 382 efficiency of a water disinfection treatment. Indeed, H₂O₂ consumption during sunlight/H₂O₂ process was relevant (no consumption in SUWWE was detected). H₂O₂ consumption in UWWE after the first phase of the sequential treatment (sunlight/H₂O₂) was 18 mg L⁻¹ and after Fe:EDDS addition it increased till 35 mg L⁻¹ at the end of the treatment. Fe precipitated during the treatment and its concentration decreased till 0.02 mM.

The aim to achieve 60% removal of the sum of total CECs was reached 60 min after the addition of Fe:EDDS (180 min of sequential treatment, $Q_{UV} = 2.5$ kJ L⁻¹). The lower efficiency of the treatment in UWWE compared to the SUWWE can be explained by the fact that UWWE usually presents higher variety of organic compounds, which can compete for the oxidative radicals generated.



392



Figure 5–Bacteria inactivation (a) and CECs degradation (b) in UWWE by sequential treatment with sunlight/H₂O₂ (H₂O₂ =50 mg L⁻¹) and SPF (120 min Fe:EDDS dosing; Fe= 0.1mM, EDDS=0.1mM) in RPR

398

In order to evaluate Fe:EDDS concentration effect in UWWE for water disinfection and 399 400 decontamination, sequential treatment was performed also using Fe:EDDS at molar ratio 1:2 (0.1:0.2 mM, data not shown) The main intention was to reinforce the photo-Fenton process 401 adding more EDDS for enhancing Fe:EDDS persistence during the process (it was degraded 402 during the whole preocess). E. coli showed similar behavior in both investigated Fe:EDDS 403 molar ratio achieving the DL (1 CFU 100 mL⁻¹) in 210 min. Salmonella spp inactivation was 404 lower in the case of Fe:EDDS 0.1:0.2 mM compared to Fe:EDDS 0.1:0.1mM. At the end of the 405 treatment, Salmonella spp concentration was 2 CFU 100 mL⁻¹ and so the DL (1 CFU 100 mL⁻ 406 ¹) was not reached. The higher organic matter concentration due to the increased EDDS dosing 407 acted as HO' scavenger, resulting in a lower inactivation efficiency (Giannakis et al., 2016; 408 Rincón and Pulgarin, 2004). CECs also showed similar behavior in both investigated Fe:EDDS 409 molar ratio. 410

In order to improve the efficiency of the sequential treatment specifically for CECs removal in 412 UWWE, sequential treatment with two dosages of Fe:EDDS (0.1:0.1 mM) was investigated. 413 First, sunlight/H₂O₂ was operated for 120 min, then, a first addition of Fe:EDDS was performed 414 415 and the SPF was run for 30 min and a second addition of Fe:EDDS (0.1:0.1 mM) was done. DL for disinfection was achieved at the end of the treatment. Regarding to the Salmonella spp, 416 the sequential treatment with two Fe:EDDS (0.1:0.1 mM) additions was less effective, 417 which can be explained by the higher DOC concentration (56.8 mg L^{-1}) due to the extra amount 418 419 of EDDS. Second addition of Fe:EDDS resulted also in a higher iron precipitation and higher turbidity (from 3.4 to 30.3NTU). It possibly resulted in higher scattering of solar radiation and 420 421 protection of bacteria against photons (García-Fernández et al., 2019).





b)

423

Figure 6–Bacteria inactivation (a) and CECs degradation (b) in UWWE by sequential treatment
in RPR by sunlight/H₂O₂– 120 min (H₂O₂=50 mg/L) and SPF with double dosage of Fe:EDDS
(120 min Fe:EDDS dosing, Fe= 0.1mM, EDDS=0.1mM and150 min second)
Fe:EDDS dosing).

428

429 The target of 60 % of CECs removal by sequential treatment with double Fe:EDDS dosage was achieved after 160 min ($Q_{UV} = 2.2 \text{ kJ/L}$), 10 min after adding the second Fe:EDDS dose (Fig. 430 6 b). CECs degradation rate was fast within 20 min after adding the first Fe:EDDS dose. Then 431 432 the reaction stopped, due to the complex destruction under natural light (Soriano-Molina et al., 2019). After adding the second Fe:EDDS dose (150 min of sequential treatment) the 433 degradation of the target CECs continued but stopped again after the same contact time. This 434 behaviour further supports the statement that the process stops as complex degradation takes 435 place. To our knowledge, sequential treatment with sunlight/H₂O₂ and SPF with double 436 437 addition of Fe:EDDS complex dose was not previously investigated, and it was found to be more effective for CECs removal compared to the other sequential treatments options discussed 438 above. However, although this approach is more effective, using two Fe:EDDS doses can result 439 in some operating problems related to the high EDDS cost and increase of DOC. In 440

particular, increased DOC concentration could limit discharge (and reuse) due to the local
legislation in force. The use of this sequential treatment option would be counterproductive for
the overall performance.

444

445 **Conclusions**

An efficient solar sequential treatment by sunlight/ H_2O_2 (50 mg L⁻¹) and SPF using EDDS (Fe 446 447 and EDDS at 0.1 mM) at circumneutral pH should be considered when inactivation of the microbial targets did not reach DL by applying only SPF in UWWE. Increase in H₂O₂ 448 concentration could improve inactivation but did not significantly affect CECs removal. 449 450 Although double dosing of Fe:EDDS complex during SPF process could result in a higher CECs removal, it cannot be considered a sustainable option. The best operation conditions for 451 disinfection and elimination of CECs in UWWE were sequential treatment with 452 sunlight/H₂O₂ (50 mg L⁻¹) during 120 min and SPF with the addition of Fe:EDDS(0.1:0.1 mM) 453 until reaching 60% elimination of the sum of total CECs. Sequential treatment is a promising 454 455 sustainable solution for tertiary treatment which allows an effective simultaneous disinfection and CECs removal for possible disposal or reuse for crop irrigation. However, future studies 456 should be performed to evaluate the best operation conditions for scaling-up the treatment, 457 458 through the support of life cycle assessment (LCA) to comprehensively assess the potential environmental impacts of the investigated treatment method compared to consolidated 459 technologies. 460

461

462 Acknowledgments

Dr. Gulnara Maniakova and prof. Luigi Rizzo would like to thank PRIMA consortium of the
European Union for the support through post-doc grant within the project "Decision supportbased approach for Sustainable Water reuse application in Agricultural Production –
DSWAP". The authors wish to thanks also the Spanish Ministry of Science and Innovation,

467 State Investigation Agency (AEI) and the European Regional Development Fund (FEDER) for

468 funding NAVIA Project (PID2019–110441RB-C32).

469

470 Disclaimer excluding European Commission and PRIMA responsibility

This manuscript reflects only the author's view and the European Commission/PRIMA are notresponsible for any use that may be made of the information it contains.

473

474 **References**

- 475 Aguas, Y., Hincapie, M., Martínez-Piernas, A.B., Agüera, A., Fernández-Ibáñez, P., Nahim-
- 476 Granados, S., Polo-López, M.I., 2019. Reclamation of Real Urban Wastewater Using

477 Solar Advanced Oxidation Processes: An Assessment of Microbial Pathogens and 74

478 Organic Microcontaminants Uptake in Lettuce and Radish. Environ. Sci. Technol. 53,

479 9705–9714. https://doi.org/10.1021/acs.est.9b00748

- 480 Cabrera-Reina, A., Miralles-Cuevas, S., Sánchez Pérez, J.A., Salazar, R., 2021. Application
- 481 of solar photo-Fenton in raceway pond reactors: A review. Science of The Total

482 Environment 800, 149653. https://doi.org/10.1016/j.scitotenv.2021.149653

- 483 Costa, E.P., Roccamante, M., Amorim, C.C., Oller, I., Sánchez Pérez, J.A., Malato, S., 2020.
- 484 New trend on open solar photoreactors to treat micropollutants by photo-Fenton at
- 485 circumneutral pH: Increasing optical pathway. Chemical Engineering Journal 385,
- 486 123982. https://doi.org/10.1016/j.cej.2019.123982

487 Delli Compagni, R., Gabrielli, M., Polesel, F., Turolla, A., Trapp, S., Vezzaro, L., Antonelli,

- 488 M., 2020. Risk assessment of contaminants of emerging concern in the context of
- 489 wastewater reuse for irrigation: An integrated modelling approach. Chemosphere 242,
- 490 125185. https://doi.org/10.1016/j.chemosphere.2019.125185
- 491 Dong, W., Sun, S.-P., Yang, X., Zhou, K., Li, Y., Wang, X., Wu, Z., Wu, W.D., Chen, X.D.,
- 492 2019. Enhanced emerging pharmaceuticals removal in wastewater after biotreatment

- 493 by a low-pressure UVA/FeIII-EDDS/H2O2 process under neutral pH conditions.
- 494 Chemical Engineering Journal 366, 539–549.
- 495 https://doi.org/10.1016/j.cej.2019.02.109
- Fent, K., Weston, A.A., Caminada, D., 2006. Ecotoxicology of human pharmaceuticals.
 Aquatic Toxicology 76, 122–159. https://doi.org/10.1016/j.aquatox.2005.09.009
- 498 García-Fernández, I., Miralles-Cuevas, S., Oller, I., Malato, S., Fernández-Ibáñez, P., Polo-
- 499 López, M.I., 2019. Inactivation of E. coli and E. faecalis by solar photo-Fenton with
- 500 EDDS complex at neutral pH in municipal wastewater effluents. Journal of Hazardous
- 501 Materials, SI: Photocatalysis:Future Trend 372, 85–93.
- 502 https://doi.org/10.1016/j.jhazmat.2018.07.037
- 503 Giannakis, S., 2018. Analogies and differences among bacterial and viral disinfection by the
- photo-Fenton process at neutral pH: a mini review. Environ Sci Pollut Res 25, 27676–
 27692. https://doi.org/10.1007/s11356-017-0926-x
- 506 Giannakis, S., Polo López, M.I., Spuhler, D., Sánchez Pérez, J.A., Fernández Ibáñez, P.,

507 Pulgarin, C., 2016. Solar disinfection is an augmentable, in situ-generated photo-

- 508 Fenton reaction—Part 1: A review of the mechanisms and the fundamental aspects of
- the process. Applied Catalysis B: Environmental 199, 199–223.
- 510 https://doi.org/10.1016/j.apcatb.2016.06.009
- 511 Giannakis, S., Rtimi, S., Pulgarin, C., 2017. Light-Assisted Advanced Oxidation Processes for
- the Elimination of Chemical and Microbiological Pollution of Wastewaters in
- 513 Developed and Developing Countries. Molecules 22, 1070.
- 514 https://doi.org/10.3390/molecules22071070
- 515 Gómez-López, V.M., Lannoo, A.-S., Gil, M.I., Allende, A., 2014. Minimum free chlorine
- residual level required for the inactivation of Escherichia coli O157:H7 and
- 517 trihalomethane generation during dynamic washing of fresh-cut spinach. Food Control
- 518 42, 132–138. https://doi.org/10.1016/j.foodcont.2014.01.034

519	Lahlou, FZ., Mackey, H.R., Al-Ansari, T., 2021. Wastewater reuse for livestock feed
520	irrigation as a sustainable practice: A socio-environmental-economic review. Journal
521	of Cleaner Production 294, 126331. https://doi.org/10.1016/j.jclepro.2021.126331
522	Leonel, L.P., Tonetti, A.L., 2021. Wastewater reuse for crop irrigation: Crop yield, soil and
523	human health implications based on giardiasis epidemiology. Science of The Total
524	Environment 775, 145833. https://doi.org/10.1016/j.scitotenv.2021.145833
525	Liao, Z., Chen, Z., Xu, A., Gao, Q., Song, K., Liu, J., Hu, HY., 2021. Wastewater treatment
526	and reuse situations and influential factors in major Asian countries. Journal of
527	Environmental Management 282, 111976.
528	https://doi.org/10.1016/j.jenvman.2021.111976
529	Malato, S., Blanco, J., Campos, A., Cáceres, J., Guillard, C., Herrmann, J.M., Fernández-
530	Alba, A.R., 2003. Effect of operating parameters on the testing of new industrial
531	titania catalysts at solar pilot plant scale. Applied Catalysis B: Environmental 42, 349-
532	357. https://doi.org/10.1016/S0926-3373(02)00270-9
533	Malato, S., Fernández-Ibáñez, P., Maldonado, M.I., Blanco, J., Gernjak, W., 2009.
534	Decontamination and disinfection of water by solar photocatalysis: Recent overview
535	and trends. Catalysis Today, MONOGRAPH: Decontamination and disinfection of
536	water by solar photocatalysis: Recent overview and trends 147, 1–59.
537	https://doi.org/10.1016/j.cattod.2009.06.018
538	Maniakova, G., Salmerón, I., Aliste, M., Inmaculada Polo-López, M., Oller, I., Malato, S.,
539	Rizzo, L., 2022. Solar photo-Fenton at circumneutral pH using Fe(III)-EDDS
540	compared to ozonation for tertiary treatment of urban wastewater: Contaminants of
541	emerging concern removal and toxicity assessment. Chemical Engineering Journal
542	431, 133474. https://doi.org/10.1016/j.cej.2021.133474
543	Maniakova, G., Salmerón, I., Nahim-Granados, S., Malato, S., Oller, I., Rizzo, L., Polo-
544	López, M.I., 2021a. Sunlight advanced oxidation processes vs ozonation for

- wastewater disinfection and safe reclamation. Science of The Total Environment 787, 545 147531. https://doi.org/10.1016/j.scitotenv.2021.147531 546 Maniakova, G., Salmerón, I., Polo-López, M.I., Oller, I., Rizzo, L., Malato, S., 2021b. 547 548 Simultaneous removal of contaminants of emerging concern and pathogens from urban wastewater by homogeneous solar driven advanced oxidation processes. 549 550 Science of The Total Environment 766, 144320. https://doi.org/10.1016/j.scitotenv.2020.144320 551 Miralles-Cuevas, S., Oller, I., Ruíz-Delgado, A., Cabrera-Reina, A., Cornejo-Ponce, L., 552 Malato, S., 2019. EDDS as complexing agent for enhancing solar advanced oxidation 553 processes in natural water: Effect of iron species and different oxidants. Journal of 554 Hazardous Materials, SI: Photocatalysis:Future Trend 372, 129–136. 555 https://doi.org/10.1016/j.jhazmat.2018.03.018 556 Moreira, N.F.F., Narciso-da-Rocha, C., Polo-López, M.I., Pastrana-Martínez, L.M., Faria, 557 J.L., Manaia, C.M., Fernández-Ibáñez, P., Nunes, O.C., Silva, A.M.T., 2018. Solar 558 559 treatment (H2O2, TiO2-P25 and GO-TiO2 photocatalysis, photo-Fenton) of organic micropollutants, human pathogen indicators, antibiotic resistant bacteria and related 560 genes in urban wastewater. Water Research 135, 195-206. 561 https://doi.org/10.1016/j.watres.2018.01.064 562 Nahim-Granados, S., Sánchez Pérez, J.A., Polo-Lopez, M.I., 2018. Effective solar processes 563 in fresh-cut wastewater disinfection: Inactivation of pathogenic E. coli O157:H7 and 564 Salmonella enteritidis. Catalysis Today, 5th European Conference on Environmental 565 Applications of Advanced Oxidation Processes 313, 79–85. 566 567 https://doi.org/10.1016/j.cattod.2017.10.042 Pignatello, J.J., Oliveros, E., MacKay, A., 2006. Advanced Oxidation Processes for Organic 568
- 569 Contaminant Destruction Based on the Fenton Reaction and Related Chemistry.

- 570 Critical Reviews in Environmental Science and Technology 36, 1–84.
- 571 https://doi.org/10.1080/10643380500326564
- 572 Rincón, A.-G., Pulgarin, C., 2004. Effect of pH, inorganic ions, organic matter and H2O2 on
- 573 E. coli K12 photocatalytic inactivation by TiO2: Implications in solar water
- disinfection. Applied Catalysis B: Environmental 51, 283–302.
- 575 https://doi.org/10.1016/j.apcatb.2004.03.007
- Rodríguez-Chueca, J., Mosteo, R., Ormad, M.P., Ovelleiro, J.L., 2012. Factorial experimental
 design applied to Escherichia coli disinfection by Fenton and photo-Fenton processes.
 Solar Energy 86, 3260–3267. https://doi.org/10.1016/j.solener.2012.08.015
- 579 Sánchez-Montes, I., Salmerón García, I., Rivas Ibañez, G., Aquino, J.M., Polo-López, M.I.,
- Malato, S., Oller, I., 2020. UVC-based advanced oxidation processes for simultaneous
 removal of microcontaminants and pathogens from simulated municipal wastewater at
- pilot plant scale. Environ. Sci.: Water Res. Technol. 6, 2553–2566.
- 583 https://doi.org/10.1039/D0EW00279H
- Scheierling, S., Bartone, C., Mara, D., Drechsel, P., 2011. Towards an agenda for improving
- 585wastewater use in agriculture. Water International 36, 420–440.
- 586 https://doi.org/10.1080/02508060.2011.594527
- 587 Soriano-Molina, P., García Sánchez, J.L., Alfano, O.M., Conte, L.O., Malato, S., Sánchez
- 588 Pérez, J.A., 2018. Mechanistic modeling of solar photo-Fenton process with Fe3+-
- 589 EDDS at neutral pH. Applied Catalysis B: Environmental 233, 234–242.
- 590 https://doi.org/10.1016/j.apcatb.2018.04.005
- 591 Soriano-Molina, P., Plaza-Bolaños, P., Lorenzo, A., Agüera, A., García Sánchez, J.L., Malato,
- 592 S., Sánchez Pérez, J.A., 2019. Assessment of solar raceway pond reactors for removal
- 593 of contaminants of emerging concern by photo-Fenton at circumneutral pH from very
- 594 different municipal wastewater effluents. Chemical Engineering Journal 366, 141–
- 595 149. https://doi.org/10.1016/j.cej.2019.02.074

- 596 Wu, X., Conkle, J.L., Ernst, F., Gan, J., 2014. Treated Wastewater Irrigation: Uptake of
- 597 Pharmaceutical and Personal Care Products by Common Vegetables under Field
 598 Conditions. Environ. Sci. Technol. 48, 11286–11293.
- 599 https://doi.org/10.1021/es502868k
- Kong, Y., Zhou, M., 2019. A critical review of the application of chelating agents to enable
- 601 Fenton and Fenton-like reactions at high pH values. Journal of Hazardous Materials
- 602 362, 436–450. https://doi.org/10.1016/j.jhazmat.2018.09.035